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(FILE 'HOME' ENTERED AT 07:59:54 ON 17 MAR 2010)

FILE 'REGISTRY' ENTERED AT 08:00:20 ON 17 MAR 2010  
ACT NGU046A/A

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L1           39 SEA SPE=ON ABB=ON PLU=ON (2627-86-3/BI OR 10420-89-0/B  
I OR 138861-14-0/BI OR 17430-98-7/BI OR 183954-15-6/BI  
OR 208848-50-4/BI OR 22526-46-1/BI OR 22526-47-2/BI OR  
27298-98-2/BI OR 2941-20-0/BI OR 3082-62-0/BI OR  
373-44-4/BI OR 3779-63-3/BI OR 3886-69-9/BI OR 402750-74-  
7/BI OR 41851-59-6/BI OR 4187-56-8/BI OR 44745-29-1/BI  
OR 45972-73-4/BI OR 467221-90-5/BI OR 475106-67-3/BI OR  
519163-49-6/BI OR 521271-41-0/BI OR 5329-79-3/BI OR  
618-36-0/BI OR 691379-96-1/BI OR 796038-32-9/BI OR  
822-06-0/BI OR 825595-86-6/BI OR 825600-99-5/BI OR  
825601-03-4/BI OR 825601-13-6/BI OR 825601-24-9/BI OR  
825635-40-3/BI OR 825635-41-4/BI OR 825635-42-5/BI OR  
825635-44-7/BI OR 83053-85-4/BI OR 99636-32-5/BI)  
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FILE 'LREGISTRY' ENTERED AT 08:00:39 ON 17 MAR 2010

L2           QUE SPE=ON ABB=ON PLU=ON C8H11N OR C9H13N C12H13N OR  
C9H13NO OR C5H13N OR C15H26N2 OR C11H15N OR C6H15N OR  
C7H17N OR C4H11NO OR C12H17N OR C10H15NO OR C11H17NO OR  
C9H13NO OR C12H17NO OR C13H19NO OR C10H15NO OR C11H17NO

FILE 'REGISTRY' ENTERED AT 08:06:54 ON 17 MAR 2010

L3           19 SEA SPE=ON ABB=ON PLU=ON L1 AND L2

FILE 'HCAPLUS' ENTERED AT 08:07:05 ON 17 MAR 2010

L4           9567 SEA SPE=ON ABB=ON PLU=ON L3

FILE 'REGISTRY' ENTERED AT 08:07:13 ON 17 MAR 2010

L5           10858 SEA SPE=ON ABB=ON PLU=ON ?DIISOCYANATE?/CNS

FILE 'HCAPLUS' ENTERED AT 08:10:01 ON 17 MAR 2010

L6           105602 SEA SPE=ON ABB=ON PLU=ON L5

L7           58 SEA SPE=ON ABB=ON PLU=ON L4 AND L6

L8           2 SEA SPE=ON ABB=ON PLU=ON L4 (L) L6

FILE 'REGISTRY' ENTERED AT 08:10:39 ON 17 MAR 2010

L9           97 SEA SPE=ON ABB=ON PLU=ON ?POLYISOCYANATE?/CNS

FILE 'HCAPLUS' ENTERED AT 08:10:58 ON 17 MAR 2010  
 L10 6116 SEA SPE=ON ABB=ON PLU=ON L9  
 L11 0 SEA SPE=ON ABB=ON PLU=ON L10 AND L4  
 L12 32677 SEA SPE=ON ABB=ON PLU=ON POLYISOCYANATE?  
 L13 5 SEA SPE=ON ABB=ON PLU=ON L4 AND L12

FILE 'REGISTRY' ENTERED AT 08:12:27 ON 17 MAR 2010  
 L14 24047 SEA SPE=ON ABB=ON PLU=ON ?ISOCYANATE?/CNS OR ?ISOCYANU  
 RATE?/CNS

FILE 'HCAPLUS' ENTERED AT 08:12:59 ON 17 MAR 2010  
 L15 180952 SEA SPE=ON ABB=ON PLU=ON L14  
 L16 649 SEA SPE=ON ABB=ON PLU=ON L4 AND L15

FILE 'ZCAPLUS' ENTERED AT 08:13:22 ON 17 MAR 2010  
 L17 QUE SPE=ON ABB=ON PLU=ON COAT?  
 L18 QUE SPE=ON ABB=ON PLU=ON RHEOLOG?

FILE 'HCAPLUS' ENTERED AT 08:13:59 ON 17 MAR 2010  
 L19 4 SEA SPE=ON ABB=ON PLU=ON L7 (L) L17  
 L20 4 SEA SPE=ON ABB=ON PLU=ON L7 (L) L18  
 L21 4 SEA SPE=ON ABB=ON PLU=ON L19 OR L20  
 L22 9 SEA SPE=ON ABB=ON PLU=ON L16 (L) L17  
 L23 4 SEA SPE=ON ABB=ON PLU=ON L16 (L) L18  
 L24 9 SEA SPE=ON ABB=ON PLU=ON L21 OR L22 OR L23  
 L25 58 SEA SPE=ON ABB=ON PLU=ON (L7 OR L16) AND L7  
 L26 4 SEA SPE=ON ABB=ON PLU=ON L25 AND L18  
 L27 5 SEA SPE=ON ABB=ON PLU=ON (L7 OR L16) (L) MOA/RL  
 L28 15 SEA SPE=ON ABB=ON PLU=ON (L4) (L) MOA/RL  
 L29 12 SEA SPE=ON ABB=ON PLU=ON L28 NOT L24

L30 7713 SEA SPE=ON ABB=ON PLU=ON L4 (L) RACT/RL  
 L31 57944 SEA SPE=ON ABB=ON PLU=ON (L6 OR L10 OR L15) (L)  
 RACT/RL  
 L32 555 SEA SPE=ON ABB=ON PLU=ON L30 AND L31  
 L33 1 SEA SPE=ON ABB=ON PLU=ON L32 AND L17  
 L34 1 SEA SPE=ON ABB=ON PLU=ON L32 AND L18  
 L35 0 SEA SPE=ON ABB=ON PLU=ON L30 (L) MOA/RACT  
 L36 0 SEA SPE=ON ABB=ON PLU=ON L30 (L) L31  
 L37 62369 SEA SPE=ON ABB=ON PLU=ON ?DIISOCYANATE?  
 L38 12 SEA SPE=ON ABB=ON PLU=ON L4 AND L37  
 L39 7 SEA SPE=ON ABB=ON PLU=ON L38 NOT (L29 OR L24)

FILE HOME

FILE REGISTRY

Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 15 MAR 2010 HIGHEST RN 1210111-73-1  
DICTIONARY FILE UPDATES: 15 MAR 2010 HIGHEST RN 1210111-73-1

New CAS Information Use Policies, enter HELP USAGETERMS for details.

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REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

<http://www.cas.org/support/stngen/stndoc/properties.html>

FILE LREGISTRY  
LREGISTRY IS A STATIC LEARNING FILE

CAS INFORMATION USE POLICIES, ENTER HELP USAGETERMS FOR DETAILS.

FILE HCAPLUS

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FILE COVERS 1907 - 17 Mar 2010 VOL 152 ISS 12  
FILE LAST UPDATED: 16 Mar 2010 (20100316/ED)  
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Dec 2009  
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Dec 2009

HCAplus now includes complete International Patent Classification (IPC) reclassification data for the first quarter of 2010.

CAS Information Use Policies apply and are available at:

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This file contains CAS Registry Numbers for easy and accurate substance identification.

#### FILE ZCAPLUS

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CAS Information Use Policies apply and are available at:

<http://www.cas.org/legal/infopolicy.html>

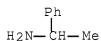
This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d 124 1-9 bib abs hitstr hitind

L24 ANSWER 1 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2007:1444461 HCAPLUS Full-text  
 DN 150:228894  
 TI Effect of single-walled carbon nanotubes on cellulose  
 phenylcarbamate chiral stationary phases  
 AU Chang, Yin-xia; Ren, Chao-xing; Ruan, Qiong; Yuan, Li-ming  
 CS Department of Chemistry, Yunnan Normal University, Kunming, 650092,  
 Peop. Rep. China  
 SO Chemical Research in Chinese Universities (2007), 23(6), 646-649  
 CODEN: CRCUED; ISSN: 1005-9040  
 PB Higher Education Press  
 DT Journal  
 LA English

AB Single-walled carbon nanotubes (SWNTs) have a high adsorption ability and nanoscale interactions. Cellulose trisphenylcarbamates possess high enantiosepn. ability in HPLC. Single-walled carbon nanotubes mixed with cellulose trisphenylcarbamate are coated on the silica gel as chiral stationary phases and higher enantiosepn. factors were obtained. After a single-walled carbon nanotube is linked to the 6-position of cellulose 2,3-bisphenylcarbamate, its enantiosepn. resolution increases compared to that of the cellulose trisphenylcarbamate. It is the 1st time that SWNTs were applied to enantiosepn. The single-walled carbon nanotubes are good promoters of chiral recognition. This method can be used to improve the enantiosepn. efficiency of the polysaccharide chiral stationary phases.

IT 618-36-0, ( $\pm$ )- $\alpha$ -Methylbenzylamine  
 2627-86-3, (-)- $\alpha$ -Methylbenzylamine 3886-69-9  
 , (+)- $\alpha$ -Methylbenzylamine  
 RL: ANT (Analyte); ANST (Analytical study)  
 (effect of single-walled carbon nanotubes on cellulose  
 phenylcarbamate chiral stationary phases for enantiosepn. by  
 HPLC)  
 RN 618-36-0 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



RN 2627-86-3 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).

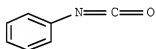


RN 3886-69-9 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



IT 103-71-9, Phenyl isocyanate, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (effect of single-walled carbon nanotubes on cellulose  
 phenylcarbamate chiral stationary phases fro enantiosepn. by  
 HPLC)  
 RN 103-71-9 HCAPLUS  
 CN Benzene, isocyanato- (CA INDEX NAME)



CC 80-4 (Organic Analytical Chemistry)  
 IT 60-18-4, L-Tyrosine, analysis 525-66-6, (±)-Propranolol  
 556-02-5, D-Tyrosine 556-03-6, Tyrosine 618-36-0,  
 (±)-α-Methylbenzylamine 1517-72-2,  
 (±)-1-(1-Naphthyl)ethanol 2627-86-3,  
 (-)-α-Methylbenzylamine 3886-69-9,  
 (+)-α-Methylbenzylamine 4199-09-1, (-)-Propranolol  
 4799-67-1, (±)-3-Benzyloxy-1,2-propanediol 5051-22-9,  
 (+)-Propranolol 15914-84-8, (-)-1-(1-Naphthyl)ethanol  
 17325-85-8, (-)-3-Benzyloxy-1,2-propanediol 32634-66-5  
 32634-68-7 42177-25-3, (+)-1-(1-Naphthyl)ethanol 56552-80-8,  
 (+)-3-Benzyloxy-1,2-propanediol 104528-81-6  
 RL: ANT (Analyte); ANST (Analytical study)  
 (effect of single-walled carbon nanotubes on cellulose  
 phenylcarbamate chiral stationary phases fro enantiosepn. by  
 HPLC)  
 IT 103-71-9, Phenyl isocyanate, reactions 9004-34-6,  
 Cellulose, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (effect of single-walled carbon nanotubes on cellulose  
 phenylcarbamate chiral stationary phases fro enantiosepn. by  
 HPLC)

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

RE.CNT 21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 2 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2007:409323 HCAPLUS Full-text

DN 148:23391

TI Study on enantioselectivity of celluloses derived by phenylcarbamate at 2, 3- or 2, 3, 6-positions

AU Chang, Yinxia; Zhou, Lingling; Yuan, Liming

CS Faculty of Chemistry and Chemical Engineering, Yunnan Normal University, Kunming, 650092, Peop. Rep. China

SO Sepu (2007), 25(2), 203-206

CODEN: SEPUER; ISSN: 1000-8713

PB Kexue Chubanshe

DT Journal

LA Chinese

AB Cellulose-2,3,6-trisphenylcarbamate, cellulose-2,3-bisphenylcarbamate, cellulose-2,3,6-tris(3,5-dimethylphenylcarbamate) and cellulose-2,3-bis(3,5-dimethylphenylcarbamate) were synthesized and resp. coated on silica gel as chiral stationary phases for HPLC. Nine pairs of enantiomers, which are (±)-phenyl-1, 2-ethanediol, (±)-2-phenyl-1-propanol, DL-α-methylbenzylamine, DL-mandelic acid, (±)-1-(1-naphthyl)ethanol, (±)-propranolol, (±)-3-benzyloxy-1,2-propanediol, DL-tyrosine and (±)-di-O,O-p-toluy-D-tartaric acid, were separated using hexane-isopropanol as mobile phase on the columns packed with the chiral stationary phases. For comparative reasons, the ratio of hexane/isopropanol in the eluent was kept at 9:1 (volume/volume) in all expts., and the chromatog. sepn. were performed at 30° with a flow rate of 0.5 mL/min. All the test solutes were detected at 254 nm. Enantiosepn. of cellulose-2, 3-bisphenylcarbamate was better than cellulose-2, 3, 6-trisphenylcarbamate for the test enantiomers, and cellulose-2, 3-bis (3, 5-dimethylphenylcarbamate) had low retention factors and short anal. times for most enantiomers and good separation factors for some racemates compared to cellulose-2, 3, 6-tris (3, 5 -dimethylphenylcarbamate).

IT 618-36-0, DL-α-Methylbenzylamine 2627-86-3

, L-α-Methylbenzylamine 3886-69-9,

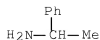
D-α-Methylbenzylamine

RL: ANT (Analyte); ANST (Analytical study)

(enantioselectivity of celluloses phenylcarbamate derivs. as chiral stationary phases for HPLC)

RN 618-36-0 HCAPLUS

CN Benzenemethanamine, α-methyl- (CA INDEX NAME)



RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



IT 103-71-9, Phenyl isocyanate, reactions

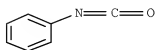
54132-75-1, 3,5-Dimethylphenyl isocyanate

RL: RCT (Reactant); RACT (Reactant or reagent)

(enantioselectivity of celluloses phenylcarbamate derivs. as  
chiral stationary phases for HPLC)

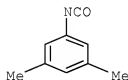
RN 103-71-9 HCAPLUS

CN Benzene, isocyanato- (CA INDEX NAME)





RN 54132-75-1 HCAPLUS  
 CN Benzene, 1-isocyanato-3,5-dimethyl- (CA INDEX NAME)



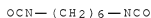
CC 80-4 (Organic Analytical Chemistry)  
 IT Silica gel, analysis  
 RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); ANST (Analytical study); USES (Uses)  
 (aminopropylsilylated, cellulose phenylcarbamate derivs. coated; enantioselectivity of celluloses phenylcarbamate derivs. as chiral stationary phases for HPLC)  
 IT 60-18-4, L-Tyrosine, analysis 90-64-2 93-56-1, (±)-Phenyl-1, 2-ethanediol 525-66-6, (±)-Propranolol 556-02-5, D-Tyrosine 556-03-6, Tyrosine 611-71-2, D-Mandelic acid 618-36-0, DL-α-Methylbenzylamine 1123-85-9, (±)-2-Phenyl-1-propanol 1517-72-2, (±)-1-(1-Naphthyl)ethanol 2627-86-3, L-α-Methylbenzylamine 3886-69-9, D-α-Methylbenzylamine 4199-09-1, (-)-Propranolol 4799-67-1 5051-22-9, (+)-Propranolol 15914-84-8, (-)-1-(1-Naphthyl)ethanol 16355-00-3, (-)-Phenyl-1, 2-ethanediol 17199-29-0, L-Mandelic acid 17325-85-8, (-)-3-Benzyloxy-1,2-propanediol 19141-40-3, (+)-2-Phenyl-1-propanol 25779-13-9, (+)-Phenyl-1, 2-ethanediol 32634-66-5 32634-68-7 37778-99-7, (-)-2-Phenyl-1-propanol 42177-25-3, (+)-1-(1-Naphthyl)ethanol 56552-80-8, (+)-3-Benzyloxy-1,2-propanediol 104528-81-6  
 RL: ANT (Analyte); ANST (Analytical study)  
 (enantioselectivity of celluloses phenylcarbamate derivs. as chiral stationary phases for HPLC)  
 IT 103-71-9, Phenyl isocyanate, reactions 9004-34-6, Cellulose, reactions 54132-75-1, 3,5-Dimethylphenyl isocyanate  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (enantioselectivity of celluloses phenylcarbamate derivs. as chiral stationary phases for HPLC)

AN 2007:13464 HCAPLUS Full-text  
 DN 146:102399  
 TI A polyurea product as thixotropic rheology modifying agent  
 IN Brinkhuis, Richard Hendrikus Gerrit  
 PA Nuplex Resins B.V., Neth.  
 SO PCT Int. Appl., 50pp.  
 CODEN: PIXXD2  
 DT Patent  
 LA English  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	WO 2007000335	A1	20070104	WO 2006-EP6250	20060628
	W:				
	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW				
	RW:				
	AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
EP	1902081	A1	20080326	EP 2006-762238	20060628
	R:				
	AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR				
JP	2008544074	T	20081204	JP 2008-518713	20060628
KR	2008031691	A	20080410	KR 2007-730468	20071227
CN	101213230	A	20080702	CN 2006-80023653	20071228
PRAI	EP 2005-105763	A	20050628		
OS	WO 2006-EP6250	W	20060628		
	MARPAT 146:102399				

AB A thixotropic agent comprising a first reaction product of a first polyisocyanate with a first a(chiral) mine and a second reaction product of a second polyisocyanate with a second amine different from the first reaction product precipitated in the presence of the colloidal particles of the first reaction product is claimed. Thus, diurea from 1,6-hexamethylene diisocyanate (I) and (R)- $\alpha$ -Methylbenzylamine (II) was prepared after I and II were formed in Setalux 1760 VB-64.

IT 822-06-0, 1,6-Hexamethylene diisocyanate  
 2627-86-3, (S)- $\alpha$ -Methylbenzylamine 3886-69-9  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (thixotropic agents comprising two reaction products from polyisocyanates and amines)  
 RN 822-06-0 HCAPLUS  
 CN Hexane, 1,6-diisocyanato- (CA INDEX NAME)



RN 2627-86-3 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 42-5 (Coatings, Inks, and Related Products)

Section cross-reference(s): 40, 41, 43, 46, 58

IT Adhesives  
 Coating materials  
 Construction materials  
 Cosmetics and personal care products  
 Detergents  
 Paper  
 Paperboard  
 Pigments, nonbiological  
 Textiles  
 Thixotropic agents  
 (thixotropic agents comprising two reaction products from polyisocyanates and amines)

IT 822-06-0, 1,6-Hexamethylene diisocyanate  
 2627-86-3, (S)- $\alpha$ -Methylbenzylamine 2885-02-1,  
 L-Alanine butyl ester 3731-52-0, 3-(Aminomethyl)pyridine  
 3886-69-9

RL: RCT (Reactant); RACT (Reactant or reagent)  
 (thixotropic agents comprising two reaction products from polyisocyanates and amines)

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

RE.CNT 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 4 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2006:700181 HCAPLUS Full-text  
 DN 145:146569  
 TI Preparation of polyurea compounds as **rheology** modifiers  
 IN Brinkhuis, Richard Hendrikus Gerrit  
 PA Nuplex Resins B.V., Neth.  
 SO PCT Int. Appl., 54 pp.  
 CODEN: PIXXD2

DT Patent  
 LA English

FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	WO 2006075000	A1	20060720	WO 2006-EP50134	20060110
W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT,				

RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT,  
 TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW  
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,  
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 EP 1679326 A1 20060712 EP 2005-75061  
 200501  
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 PL, SK, BA, HR, IS, YU  
 EP 1846469 A1 20071024 EP 2006-700743  
 200601  
 10  
 R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,  
 IE, IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK,  
 TR, AL, BA, HR, MK, YU  
 CN 101103060 A 20080109 CN 2006-80002086  
 200601  
 10  
 JP 2008526922 T 20080724 JP 2007-550788  
 200601  
 10  
 KR 2007100947 A 20071015 KR 2007-716675  
 200707  
 20  
 US 20080139755 A1 20080612 US 2007-795095  
 200710  
 30  
 PRAI EP 2005-75061 A 20050111  
 US 2005-654455P P 20050222  
 WO 2006-EP50134 W 20060110

## ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB Polyurea compds. are obtainable by reacting one or more polyisocyanates with one or more non-chiral mono-amines (I) and one or more chiral mono-amines (II) and co-precipitating the reaction products to form the polyurea compound, wherein 2 - 98 mol % of the mono-amines in the polyurea compound are chiral mono-amines. The invention also relates to the use of said polyurea compound as a rheol. modification agent, in particular as sag control agent (SCA) in coating compns. The invention further relates to sag control agent compns., coating compns. and coatings comprising the polyurea compound as sag control agent.

IT 822-06-ODP, 1,6-Hexamethylene-diisocyanate, reaction products with (+/-)-a-methylbenzylamine and benzylamine

2627-86-3DP, S-(-)- $\alpha$ -Methylbenzylamine, reaction products with R-(+)- $\alpha$ -methylbenzylamine and benzylamine and hexamethylenediisocyanate 3886-69-9DP,

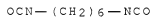
R-(+)- $\alpha$ -Methylbenzylamine, reaction products with S-(-)- $\alpha$ -methylbenzylamine and benzylamine and hexamethylenediisocyanate

RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP (Preparation); USES (Uses)

(preparation of polyurea compds. as rheol. modifiers)

RN 822-06-0 HCAPLUS

CN Hexane, 1,6-diisocyanato- (CA INDEX NAME)



RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 37-3 (Plastics Manufacture and Processing)

ST polyurea rheol modifier

IT Plastics, uses

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

- (engineering; preparation of polyurea compds. as rheol. modifiers)
- IT Ureas  
 RL: IMF (Industrial manufacture); MOA (Modifier or additive use);  
 PREP (Preparation); USES (Uses)  
 (poly-; preparation of polyurea compds. as rheol. modifiers)
- IT Adhesives  
 Coating materials  
 Construction materials  
 Cosmetics  
 Detergents  
 Drilling fluids  
 Paper  
 Paperboard  
 Pigments, nonbiological  
 Textiles  
 (preparation of polyurea compds. as rheol. modifiers)
- IT Inks  
 (printing; preparation of polyurea compds. as rheol. modifiers)
- IT 100-46-9DP, Benzylamine, reaction products with  
 (+/-)-a-methylbenzylamine and 1,6-hexamethylene-diisocyanate  
 100-46-9DP, Benzylamine, reaction products with L-alanine butylester  
 and 1,6-hexamethylene-diisocyanate 100-46-9DP, Benzylamine,  
 reaction products with R-(+)-a-methylbenzylamine and  
 S-(-)-a-methylbenzylamine and hexamethylenediisocyanate  
 822-06-0DP, 1,6-Hexamethylene-diisocyanate, reaction  
 products with (+/-)-a-methylbenzylamine and benzylamine  
 822-06-0DP, 1,6-Hexamethylene-diisocyanate, reaction  
 products with L-alanine butylester and benzylamine  
 822-06-0DP, Hexamethylenediisocyanate, reaction products  
 with R-(+)-a-methylbenzylamine and S-(-)-a-methylbenzylamine and  
 benzylamine 2627-86-3DP,  
 S-(-)-a-Methylbenzylamine, reaction products with  
 R-(+)-a-methylbenzylamine and benzylamine and  
 hexamethylenediisocyanate 2885-02-1DP, L-Alanine butylester,  
 reaction products with benzylamine and  
 1,6-hexamethylene-diisocyanate 3886-69-9DP,  
 R-(+)-a-Methylbenzylamine, reaction products with  
 S-(-)-a-methylbenzylamine and benzylamine and  
 hexamethylenediisocyanate 3886-69-9DP,  
 (+/-)-a-Methylbenzylamine, reaction products with benzylamine  
 and 1,6-hexamethylene-diisocyanate  
 RL: IMF (Industrial manufacture); MOA (Modifier or additive use);  
 PREP (Preparation); USES (Uses)  
 (preparation of polyurea compds. as rheol. modifiers)
- IT 475106-67-3, Setal 1715VX74 519163-49-6, Setalux 1767VV65

521271-41-0, Setal 166SS80 691379-94-9, Setalux 1760VB64  
 691379-95-0, Setalux 1770VS70 825635-41-4, Setalux 1757VV70  
 825635-44-7, Setalux 8503SS60  
 RL: POF (Polymer in formulation); TEM (Technical or engineered  
 material use); USES (Uses)

(preparation of polyurea compds. as rheol. modifiers)

OSC.G 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2  
 CITINGS)

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 5 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2006:699881 HCAPLUS Full-text

DN 145:169063

TI Commixtures for use in rheology modification of  
 coating binders

IN Brinkhuis, Richard Hendrikus Gerrit; Bosma, Martin

PA Nuplex Resins B.V., Neth.

SO PCT Int. Appl., 36 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 2

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
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PI WO 2006074895	A1	20060720	WO 2006-EP138	

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,  
 CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,  
 GB, GD, GE, GR, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM,  
 KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG,  
 MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT,  
 RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT,  
 TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW  
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,  
 IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR,  
 BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD,  
 TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,  
 ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

EP 1679326

A1

20060712

EP 2005-75061

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC,  
 PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU,  
 PL, SK, BA, HR, IS, YU



EP 1838747	A1	20071003	EP 2006-702653	200601 10
R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR				
CN 101103061	A	20080109	CN 2006-80002090	200601 10
JP 2008527126	T	20080724	JP 2007-550741	200601 10
KR 2007097079	A	20071002	KR 2007-717148	200707 25
US 20080146720	A1	20080619	US 2007-795096	200710 30

PRAI EP 2005-75061 A 20050111  
 EP 2005-105754 A 20050628  
 WO 2006-EP138 W 20060110

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB A commixt. is used as a sag control agent (SCA) in a composition to be cured at a temperature (T<sub>cur</sub>) >60°, the composition comprising a binder and commixt., where the commixt. comprises (a) a thixotropy-inducing particulate polyurea product having a melting temperature (T<sub>m1</sub>) ≥10° below the intended curing temperature, satisfying the requirement T<sub>m1</sub> < (T<sub>cur</sub> - 10°), and (b) a second thixotropy-inducing particulate component that retains its particulate nature at temps. at least up to the curing temperature

IT 822-06-0D, 1,6-Hexamethylene diisocyanate, urea adduct with amine 2627-86-3D, S-(-)-α-Methylbenzylamine, urea adduct with HDI 3886-69-9D, urea adduct with HDI RL: MOA (Modifier or additive use); USES (Uses) (commixts. of polyurea/particles for sag control of coating binders)

RN 822-06-0 HCAPLUS

CN Hexane, 1,6-diisocyanato- (CA INDEX NAME)

OCN—(CH<sub>2</sub>)<sub>6</sub>—NCO

RN 2627-86-3 HCAPLUS

CN Benzenemethanamine, α-methyl-, (αS)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 42-5 (Coatings, Inks, and Related Products)

ST polyurea particle sag control agent **coating**

IT **Coating materials**

Thixotropic agents

(commixts. of polyurea/particles for sag control of **coating binders**)

IT Polyureas

RL: MOA (Modifier or additive use); USES (Uses)

(commixts. of polyurea/particles for sag control of **coating binders**)

IT 899821-42-2P 899821-43-3P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(**coating**; commixts. of polyurea/particles for sag control of **coating binders**)

IT 822-06-0D, 1,6-Hexamethylene diisocyanate, urea adduct

with amine 2627-86-3D, S-(-)- $\alpha$ -Methylbenzylamine,

urea adduct with HDI 2885-02-1D, L-Alanine butyl ester, urea

adduct with HDI 3886-69-9D, urea adduct with HDI

5332-73-0D, 3-Methoxypropylamine, urea adduct with HDI

882169-71-3, Setalux 91756 900181-60-4, Setalux 91795

RL: MOA (Modifier or additive use); USES (Uses)

(commixts. of polyurea/particles for sag control of **coating binders**)

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1

## CITINGS)

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 6 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN  
AN 2006:117175 HCAPLUS Full-text  
DN 144:192639  
TI Substituted organopolysiloxanes and use thereof  
IN Wilson, John Robert Howe; Sullivan, Alice Caroline; Man, Siud Pui  
PA Phosphonics Ltd., UK  
SO PCT Int. Appl., 45 pp.  
CODEN: PIXXD2  
DT Patent  
LA English  
FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	----	-----	-----	
PI WO 2006013060	A1	20060209	WO 2005-EP8189	200507 26
W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW			
RW:	AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM			
EP 1786850	A1	20070523	EP 2005-775020	200507 26
EP 1786850	B1	20081126		
R:	AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR			
CN 101023120	A	20070822	CN 2005-80031451	200507 26
JP 2008508406	T	20080321	JP 2007-524235	200507 26
AT 415438	T	20081215	AT 2005-775020	

200507  
26

IN 2007DN00941 A 20070803 IN 2007-DN941

200702  
05

US 20090098082 A1 20090416 US 2008-659329

200810  
28

PRAI GB 2004-17345 A 20040804  
GB 2004-26622 A 20041204  
WO 2005-EP8189 W 20050726

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB Products having combinations of siloxane, silsesquioxane, and silicate units and S-containing organic substituents, optionally, metals, and, optionally, crosslinking groups containing Si, Al, Ti, or other oxo metal bridging systems are manufactured for use as scavengers for the removal of unwanted organic and inorg. compds., for solid phase extraction, for solid phase synthesis, for acid and metal mediated heterogeneous catalysis, for metal ion abstraction and for the immobilization of bio-mols. A typical product was manufactured by reaction of 1.02 mol trimethoxyvinylsilane at 115° with 0.97 mol Me thioglycolate in the presence of di-tert-Bu peroxide and hydrolytic polymerization of 38.1 g intermediate with 62.4 g tetra-Et orthosilicate at 80° in a mixture containing 200 mL MeOH and 36 mL 1 M HCl.

IT 3886-69-9DP, (+)- $\alpha$ -Methylbenzylamine, reaction products with silica and Me [(trimethoxysilyl)ethyl]thioglycolate  
RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
(organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units)

RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).

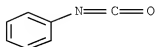


IT 103-71-9, Phenyl isocyanate, processes  
RL: REM (Removal or disposal); PROC (Process)  
(organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units for

removal of metals and organic compds. from liquid media)

RN 103-71-9 HCAPLUS

CN Benzene, isocyanato- (CA INDEX NAME)



CC 35-6 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 9, 21, 60

IT Coating materials

Medical goods

(organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units for coatings for medical devices)

IT Coating materials

(water-resistant; organic sulfur-containing group-substituted products

having combinations of siloxane, silsesquioxane, and silicate units for waterproof coatings)

IT 60-24-2DP, 2-Mercaptoethanol, reaction products with silica and butanol 71-36-3DP, 1-Butanol, reaction products

Mercaptoethanol-modified silica 109-01-3DP, 1-Methylpiperazine, reaction products Mercaptoethanol-modified silica 109-55-7DP, 3-Dimethylaminopropylamine, reaction products with Me

[(trimethoxysilyl)ethyl]thioglycolate-tetraethyl orthosilicate copolymer 111-40-0DP, Diethylenetriamine, reaction products with Me

[(trimethoxysilyl)ethyl]thioglycolate-tetraethyl orthosilicate copolymer 112-24-3DP, reaction products with Et

[(trimethoxysilyl)ethyl]thioglycolate and silica 112-57-2DP, reaction products with silica and Me

[(trimethoxysilyl)ethyl]thioglycolate 140-31-8DP, 1-(2-Aminoethyl)piperazine, reaction products with silica and Me

[(trimethoxysilyl)ethyl]thioglycolate 302-01-2DP, Hydrazine, reaction products with Me [(trimethoxysilyl)ethyl]thioglycolate-

tetraethyl orthosilicate copolymer 1344-09-8DP, Sodium silicate, reaction products with Me [(trimethoxysilyl)ethyl]thioglycolate-

tetraethyl orthosilicate copolymer 1344-28-1DP, Alumina, reaction products with Me [(trimethoxysilyl)ethyl]thioglycolate

3731-52-0DP, 3-(Aminomethyl)pyridine, reaction products with silica and Me [(trimethoxysilyl)ethyl]thioglycolate 3886-69-9DP

, (+)- $\alpha$ -Methylbenzylamine, reaction products with silica and Me [(trimethoxysilyl)ethyl]thioglycolate 5332-73-0DP,

3-Methoxypropylamine, reaction products with Me [(trimethoxysilyl)ethyl]thioglycolate-tetraethyl orthosilicate copolymer 7439-96-5DP, Manganese, complexes with organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units 7440-02-0DP, Nickel, complexes with organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units 7440-05-3DP, Palladium, complexes with hydrolyzed Me [(trimethoxysilyl)ethyl]thioglycolate-tetraethyl orthosilicate copolymer 7440-06-4DP, Platinum, complexes with organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units 7440-16-6DP, Rhodium, complexes with organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units 7440-48-4DP, Cobalt, complexes with organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units 7440-50-8DP, Copper, complexes with Me [(trimethoxysilyl)ethyl]thioglycolate-tetraethyl orthosilicate copolymer 7631-86-9DP, Silica, reaction products with Et [(trimethoxysilyl)ethyl]thioglycolate and triethylenetetraamine 35320-23-1DP, (-)-2-Amino-1-propanol, reaction products with silica and Me [(trimethoxysilyl)ethyl]thioglycolate 39660-55-4DP, Octafluoropentanol, reaction products Mercaptoethanol-modified silica 70615-97-3DP, reaction products with silica and amines 111597-50-3DP, reaction products with silica 875121-66-7DP, derivs. 875121-66-7P, Methyl [2-(trimethoxysilyl)ethyl]thioglycolate-tetraethyl orthosilicate copolymer 875121-67-8DP, Ethyl [2-(trimethoxysilyl)ethyl]thioglycolate, reaction products with silica and triethylenetetraamine 875121-71-4P, 1,3-Bis[[2-(trimethoxysilyl)ethyl]thio]propane-3-mercaptopropyl 2-(trimethoxysilyl)ethyl sulfide-tetraethyl orthosilicate copolymer 875289-32-0P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(organic sulfur-containing group-substituted products having combinations of siloxane, silsesquioxane, and silicate units)

IT 98-88-4, Benzoyl chloride 100-46-9, Benzylamine, processes 100-52-7, Benzaldehyde, processes 103-71-9, Phenyl isocyanate, processes 104-15-4, p-Toluenesulfonic acid, processes 111-26-2, Hexylamine 541-41-3, Ethyl chloroformate 3375-31-3 7440-02-0, Nickel, processes 7440-05-3, Palladium, processes 7440-06-4, Platinum, processes 7440-15-5, Rhodium, processes 7440-16-6, Rhodium, processes 7447-39-4, Cupric chloride,

processes 7705-08-0, Ferric chloride, processes 7761-88-8,  
 Silver nitrate, processes 10025-99-7, Potassium chloroplatinate  
 10049-07-7, Rhodium chloride 13007-90-4,  
 Bis(triphenylphosphine)dicarbonylnickel 13965-03-2,  
 Bis(triphenylphosphine)palladium chloride 14221-01-3,  
 Tetrakis(triphenylphosphine)palladium 14694-95-2,  
 Chlorotris(triphenylphosphine)rhodium

RL: REM (Removal or disposal); PROC (Process)

(organic sulfur-containing group-substituted products having  
 combinations of siloxane, silsesquioxane, and silicate units for  
 removal of metals and organic compds. from liquid media)

OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3  
 CITINGS)

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 7 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2005:58292 HCAPLUS Full-text

DN 142:136649

TI Carbon-substituted methyl amine derivatives and their use as a  
 rheology control agent for coating compositions

IN Brinkhuis, Richard Hendrikus Gerrit; Venderbosch, Rudolf Anthonius  
 Maria

PA Akzo Nobel N.V., Neth.

SO PCT Int. Appl., 43 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 2

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	WO 2005005557	A1	20050120	WO 2004-EP7597
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200407  
08

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,  
 CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,  
 GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP,  
 KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW,  
 MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD,  
 SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ,  
 VC, VN, YU, ZA, ZM, ZW

RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW,  
 AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ,  
 DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL,  
 PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ,  
 GW, ML, MR, NE, SN, TD, TG

EP 1641887	A1	20060405	EP 2004-740876	20040708
CN 1816598	A	20060809	CN 2004-80019232	20040708
CN 100457837	C	20090204		
BR 2004012310	A	20060822	BR 2004-12310	20040708
JP 2009513739	T	20090402	JP 2006-518166	20040708
KR 2006086931	A	20060801	KR 2006-700442	20060107
US 20060289828	A1	20061228	US 2006-564046	20060510
PRAI EP 2003-77152	A	20030708		
US 2003-530240P	P	20031218		
WO 2004-EP7597	W	20040708		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS MARPAT 142:136649

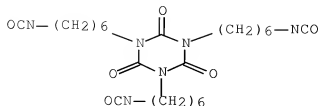
AB The invention relates to the use as sagging controlling agent (SCA) in coating compns. of rheol. control agents obtainable by reacting one or more polyisocyanates with one or more optically active amines or by reacting one or more polyamines with one or more optically active isocyanates. The invention also relates to rheol. control agents obtainable as described above using specific polyisocyanates or polyamines. In addition the invention relates to the use of these rheol. control agents in various applications.

IT 3779-63-3, Hexamethylene diisocyanate isocyanurate  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (crosslinker; carbon-substituted Me amine derivs. use as rheol. control agent for coating compns.)

RN 3779-63-3 HCAPLUS

CN 1,3,5-Triazine-2,4,6(1H,3H,5H)-trione,  
 1,3,5-tris(6-isocyanatohexyl)- (CA INDEX NAME)



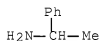


IT 618-36-0, DL-1-Phenylethylamine 822-06-0D,  
 Hexamethylene diisocyanate, reaction products with  
 L-1-amino-ethylbenzene 2627-86-3,  
 S-(-)- $\alpha$ -Methylbenzylamine 2627-86-3D,  
 L.- $\alpha$ -Methylbenzylamine, reaction products with hexamethylene  
 diisocyanate 3886-69-9D, D.- $\alpha$ -Methylbenzylamine,  
 reaction products with hexamethylene diisocyanate  
 5329-79-3, 2-Aminohexane 22526-46-1  
 22526-47-2 41851-59-6,  
 (S)-(-)-1-(4-Methoxyphenyl)ethylamine 44745-29-1  
 45972-73-4 83053-85-4 99636-32-5  
 183954-15-6 208848-50-4 402750-74-7  
 796038-32-9 825600-99-5 825601-03-4  
 825601-13-6 825601-24-9

RL: MOA (Modifier or additive use); USES (Uses)  
 (rheol. control agent; carbon-substituted Me amine  
 derivs. use as rheol. control agent for coating  
 compns.)

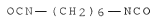
RN 618-36-0 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



RN 822-06-0 HCAPLUS

CN Hexane, 1,6-diisocyanato- (CA INDEX NAME)



RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

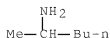
CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



RN 5329-79-3 HCAPLUS

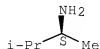
CN 2-Hexanamine (CA INDEX NAME)



RN 22526-46-1 HCAPLUS

CN 2-Butanamine, 3-methyl-, (2S)- (CA INDEX NAME)

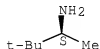
Absolute stereochemistry. Rotation (-).



RN 22526-47-2 HCAPLUS

CN 2-Butanamine, 3,3-dimethyl-, (2S)- (CA INDEX NAME)

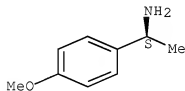
Absolute stereochemistry. Rotation (+).



RN 41851-59-6 HCAPLUS

CN Benzenemethanamine, 4-methoxy- $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

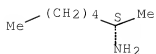
Absolute stereochemistry. Rotation (-).



RN 44745-29-1 HCAPLUS

CN 2-Heptanamine, (2S)- (CA INDEX NAME)

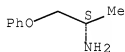
Absolute stereochemistry. Rotation (+).



RN 45972-73-4 HCAPLUS

CN 2-Propanamine, 1-phenoxy-, (2S)- (CA INDEX NAME)

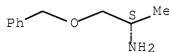
Absolute stereochemistry. Rotation (+).



RN 83053-85-4 HCAPLUS

CN 2-Propanamine, 1-(phenylmethoxy)-, (2S)- (CA INDEX NAME)

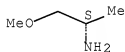
Absolute stereochemistry. Rotation (+).



RN 99636-32-5 HCAPLUS

CN 2-Propanamine, 1-methoxy-, (2S)- (CA INDEX NAME)

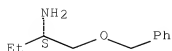
Absolute stereochemistry. Rotation (-).



RN 183954-15-6 HCAPLUS

CN 2-Butanamine, 1-(phenylmethoxy)-, (2S)- (CA INDEX NAME)

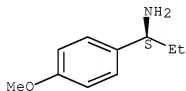
Absolute stereochemistry.



RN 208848-50-4 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -ethyl-4-methoxy-, ( $\alpha$ S)- (CA INDEX NAME)

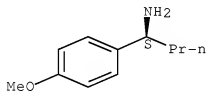
Absolute stereochemistry. Rotation (-).



RN 402750-74-7 HCAPLUS

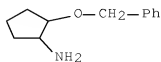
CN Benzenemethanamine, 4-methoxy- $\alpha$ -propyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry.



RN 796038-32-9 HCAPLUS

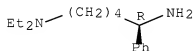
CN Cyclopentanamine, 2-(phenylmethoxy)- (CA INDEX NAME)



RN 825600-99-5 HCAPLUS

CN 1,5-Pentanediamine, N5,N5-diethyl-1-phenyl-, (1R)- (CA INDEX NAME)

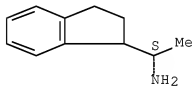
Absolute stereochemistry.



RN 825601-03-4 HCAPLUS

CN 1H-Indene-1-methanamine, 2,3-dihydro- $\alpha$ -methyl-, ( $\alpha$ S)-  
(CA INDEX NAME)

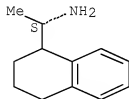
Absolute stereochemistry.



RN 825601-13-6 HCAPLUS

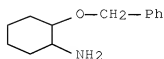
CN 1-Naphthalenemethanamine, 1,2,3,4-tetrahydro- $\alpha$ -methyl-,  
( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry.



RN 825601-24-9 HCAPLUS

CN Cyclohexanamine, 2-(phenylmethoxy)- (CA INDEX NAME)



IC ICM C09D005-04  
ICS C08G018-28; C07C273-18

CC 42-5 (Coatings, Inks, and Related Products)

ST carbon substituted methyl amine deriv **rheol** control agent  
**coating**

IT Adhesives  
Carpets  
Coating materials  
Cosmetics  
Detergents  
Leather  
Mining  
Paper  
Paperboard  
Pigments, nonbiological  
Textiles  
(carbon-substituted Me amine derivs. use as **rheol**.  
control agent for **coating** compns.)

IT Detergents  
(cleaning compns.; carbon-substituted Me amine derivs. use as  
**rheol**. control agent for **coating** compns.)

IT Acrylic polymers, uses  
RL: POF (Polymer in formulation); TEM (Technical or engineered  
material use); USES (Uses)  
(hydroxy-containing; carbon-substituted Me amine derivs. use as  
**rheol**. control agent for **coating** compns.)

IT Polyesters, uses  
RL: POF (Polymer in formulation); TEM (Technical or engineered  
material use); USES (Uses)  
(hydroxy-terminated; carbon-substituted Me amine derivs. use as  
**rheol**. control agent for **coating** compns.)

IT Inks  
(printing; carbon-substituted Me amine derivs. use as  
**rheol**. control agent for **coating** compns.)

IT 467221-90-5, Setalux 1767 475106-67-3, Setal 1715VX74  
519163-49-6, Setalux 1767VV65 521271-41-0, Setal 166SS80

691379-96-1, Setalux 1795VX74 825595-86-6 825635-40-3, Setalux  
 1198SS70 825635-41-4, Setalux 1757VV70 825635-42-5, Setalux 1770  
 825635-44-7, Setalux 8503SS60  
 RL: POF (Polymer in formulation); TEM (Technical or engineered  
 material use); USES (Uses)  
 (carbon-substituted Me amine derivs. use as rheol.  
 control agent for coating compns.)

IT 3779-63-3, Hexamethylene diisocyanate isocyanurate  
 138861-14-0, Tolonate HDT-LV  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (crosslinker; carbon-substituted Me amine derivs. use as  
 rheol. control agent for coating compns.)

IT 373-44-4, 1,8-Diaminooctane 618-36-0,  
 DL-1-Phenylethylamine 822-06-0D, Hexamethylene  
 diisocyanate, reaction products with L-1-amino-ethylbenzene  
 2627-86-3, S-(-)- $\alpha$ -Methylbenzylamine  
 2627-86-3D, L- $\alpha$ -Methylbenzylamine, reaction products  
 with hexamethylene diisocyanate 2941-20-0, 1-Phenylpropylamine  
 3082-62-0 3886-69-9D, D- $\alpha$ -Methylbenzylamine,  
 reaction products with hexamethylene diisocyanate 4187-56-8,  
 (S)-4-Chloro- $\alpha$ -methylbenzenemethanamine 5329-79-3,  
 2-Aminohexane 10420-89-0, S-1-(1-Naphthyl)ethylamine 17430-98-7,  
 (S)-(+)-1-Cyclohexylethylamine 22526-46-1  
 22526-47-2 27298-98-2 41851-59-6,  
 (S)-(-)-1-(4-Methoxyphenyl)ethylamine 44745-29-1  
 45972-73-4 83053-85-4 99636-32-5  
 183954-15-6 208848-50-4 402750-74-7  
 796038-32-9 825600-99-5 825601-03-4  
 825601-13-6 825601-24-9  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (rheol. control agent; carbon-substituted Me amine  
 derivs. use as rheol. control agent for coating  
 compns.)

OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3  
 CITINGS)

RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 8 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2003:715111 HCAPLUS Full-text  
 DN 139:390336  
 TI Characterization of derivatized cellulose coated zirconia  
 as chiral stationary phase by high-performance liquid chromatography  
 AU Dun, Huijuan; Han, Xiaolian; Liu, Chunhui; Li, Yongmin; Chen, Liren  
 CS Lanzhou Institute of Chemical Physics, Chinese Academy of Sciences,  
 Lanzhou, 730000, Peop. Rep. China  
 SO Fenxi Huaxue (2003), 31(8), 901-905



CODEN: FHHHDT; ISSN: 0253-3820

PB Kexue Chubanshe

DT Journal

LA Chinese

AB The paper describes a procedure for preparing cellulose tris(3,5-dimethylphenylcarbamate) coated zirconia (CDMPC- coated ZrO<sub>2</sub>). Some neutral, basic and acidic enantiomers were enantiosepd. under normal-phase conditions by HPLC. Surface basic property of zirconia has profound influences on retention, selectivity and resolution of different racemate mixts. Good enantiomeric resolns. for neutral and basic analytes were achieved, while acidic enantiomer did not elute from the column unless an acidic additive was presented in mobile phase.

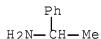
IT 618-36-0, (+)- $\alpha$ -Phenylethylamine  
2627-86-3, (-)- $\alpha$ -Phenylethylamine 3886-69-9

, (+)- $\alpha$ -Phenylethylamine

RL: ANT (Analyte); ANST (Analytical study)

(analyte; preparation and use of derivatized cellulose coated zirconia as chiral stationary phase by HPLC)

RN 618-36-0 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)

RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



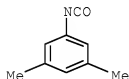
RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



IT 54132-75-1, 3,5-Dimethylphenyl isocyanate  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (in preparation of derivatized cellulose coated zirconia as  
 chiral stationary phase by HPLC)  
 RN 54132-75-1 HCAPLUS  
 CN Benzene, 1-isocyanato-3,5-dimethyl- (CA INDEX NAME)



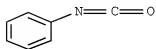
CC 80-4 (Organic Analytical Chemistry)  
 ST cellulose dimethylphenylcarbamate coated zirconia chiral  
 stationary phase HPLC  
 IT HPLC stationary phases  
 (chiral; preparation and use of derivatized cellulose coated  
 zirconia as chiral stationary phase by HPLC)  
 IT Resolution (separation)  
 (chromatog.; preparation and use of derivatized cellulose  
 coated zirconia as chiral stationary phase by HPLC)  
 IT 93-54-9, (±)-α-Phenylpropanol 98-85-1,  
 (±)-α-Phenylethanol 613-87-6, (-)-α-Phenylpropanol  
 618-36-0, (±)-α-Phenylethylamine 1445-91-6,  
 (-)-α-Phenylethanol 1517-69-7, (+)-α-Phenylethanol  
 1565-74-8, (+)-α-Phenylpropanol 2627-86-3,  
 (-)-α-Phenylethylamine 3886-69-9,  
 (+)-α-Phenylethylamine 30012-51-2, (±)-Naproxen methyl  
 ester  
 RL: ANT (Analyte); ANST (Analytical study)  
 (analyte; preparation and use of derivatized cellulose coated  
 zirconia as chiral stationary phase by HPLC)  
 IT 9004-34-6, Cellulose, reactions 54132-75-1,

- 3,5-Dimethylphenyl isocyanate  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (in preparation of derivatized cellulose coated zirconia as  
 chiral stationary phase by HPLC)
- IT 22204-53-1, (+)-Naproxen 23979-41-1, (-)-Naproxen 23981-80-8,  
 (±)-Naproxen 26159-35-3, (+)-Naproxen methyl ester  
 31220-35-6, (+)-Naproxen ethyl ester 37961-57-2, (±)-Naproxen  
 ethyl ester 81623-44-1, (-)-Naproxen methyl ester 84890-25-5,  
 (-)-Naproxen ethyl ester 105052-64-0 124649-62-3 181231-10-7  
 RL: ANT (Analyte); ANST (Analytical study)  
 (preparation and use of derivatized cellulose coated  
 zirconia as chiral stationary phase by HPLC)
- IT 1314-23-4, Zirconia, analysis  
 RL: ARU (Analytical role, unclassified); NUU (Other use,  
 unclassified); ANST (Analytical study); USES (Uses)  
 (preparation and use of derivatized cellulose coated  
 zirconia as chiral stationary phase by HPLC)
- IT 103938-44-9P, Cellulose tris(3,5-dimethylphenylcarbamate)  
 RL: ARU (Analytical role, unclassified); NUU (Other use,  
 unclassified); PRP (Properties); SPN (Synthetic preparation); ANST  
 (Analytical study); PREP (Preparation); USES (Uses)  
 (preparation and use of derivatized cellulose coated  
 zirconia as chiral stationary phase by HPLC)
- L24 ANSWER 9 OF 9 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2001:122244 HCAPLUS Full-text  
 DN 134:304782  
 TI Optical resolution on amylose-tris(phenylcarbamate) chiral  
 stationary phase  
 AU Liu, Yue-qi; Zhou, Wen-feng; Han, Xiao-qian; Jiang, Sheng-xiang;  
 Chen, Li-ren  
 CS (Lanzhou Institute of Chemical Physics, The Chinese Academy of  
 Sciences, Lanzhou, 730000, Peop. Rep. China  
 SO Fenxi Ceshi Xuebao (2001), 20(1), 43-45  
 CODEN: FCEXES; ISSN: 1004-4957  
 PB Fenxi Ceshi Xuebao Bianjibu  
 DT Journal  
 LA Chinese  
 AB A chiral stationary phase was prepared by coating amylose-  
 tris(phenylcarbamate) (ATPC) onto aminopropylated silica gel.  
 Optical resolution of a range of racemic compds. was studied. The  
 structural character of the samples that influences chiral  
 recognition is discussed. A model of interaction between the  
 stationary phase and the samples was presented.
- IT 103-71-9, Phenyl isocyanate, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (in preparation of amylose-tris(phenylcarbamate) coated

aminopropylated silica gel chiral stationary phase for liquid chromatog.)

RN 103-71-9 HCAPLUS

CN Benzene, isocyanato- (CA INDEX NAME)



IT 618-36-0 2627-86-3 3886-69-9

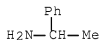
RL: ANT (Analyte); PEP (Physical, engineering or chemical process);

ANST (Analytical study); PROC (Process)

(optical resolution by liquid chromatog. on amylose-tris(phenylcarbamate) coated aminopropylated silica gel chiral stationary phase)

RN 618-36-0 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



- CC 80-4 (Organic Analytical Chemistry)  
Section cross-reference(s): 66
- IT Resolution (separation)  
(chromatog.; optical resolution by liquid chromatog. on amylose-tris(phenylcarbamate) **coated** aminopropylated silica gel chiral stationary phase)
- IT Particle size  
Pore size  
Surface area  
(of amylose-tris(phenylcarbamate) **coated** aminopropylated silica gel chiral stationary phase for liquid chromatog.)
- IT Silica gel, analysis  
RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); PRP (Properties); ANST (Analytical study); USES (Uses)  
(reaction products; optical resolution by liquid chromatog. on amylose-tris(phenylcarbamate) **coated** aminopropylated silica gel chiral stationary phase)
- IT 103-71-9, Phenyl isocyanate, reactions 919-30-2,  
γ-Aminopropyltriethoxysilane 9005-82-7, Amylose  
RL: RCT (Reactant); RACT (Reactant or reagent)  
(in preparation of amylose-tris(phenylcarbamate) **coated** aminopropylated silica gel chiral stationary phase for liquid chromatog.)
- IT 93-54-9 98-85-1 119-53-9 613-87-6 **618-36-0**  
698-87-3 1445-91-6 1517-68-6 1517-69-7 1565-74-8 1572-95-8  
**2627-86-3 3886-69-9** 5349-60-0 5928-66-5  
5928-67-6 7452-01-9 13856-85-4 42052-51-7 69897-46-7  
72237-27-5 73854-04-3 73890-73-0 105836-13-3 105836-14-4  
110611-21-7 114389-71-8  
RL: ANT (Analyte); PEP (Physical, engineering or chemical process);  
ANST (Analytical study); PROC (Process)  
(optical resolution by liquid chromatog. on amylose-tris(phenylcarbamate) **coated** aminopropylated silica gel chiral stationary phase)
- IT 9047-05-6, Amylose-tris(phenylcarbamate)  
RL: ARU (Analytical role, unclassified); NUU (Other use, unclassified); PRP (Properties); ANST (Analytical study); USES (Uses)

(optical resolution by liquid chromatog. on  
amylose-tris(phenylcarbamate) coated aminopropylated  
silica gel chiral stationary phase)

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1  
CITINGS)

=> d 139 1-7 bib abs hitstr hitind

L39 ANSWER 1 OF 7 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2008:1071619 HCAPLUS Full-text

DN 150:135752

TI Preparation and enantioseparation of polymer-type chiral stationary  
phases derived from (1S,2R)-(+)-2-amino-1,2-diphenylethanol

AU Huang, Shao-Hua; Zhang, Jun-Yi; Li, Shi-Rong; Yin, Chuan-Qi; Pan,  
Zhi-Quan; Bai, Zheng-Wu

CS Key Laboratory of Green Chemical Process of Ministry of Education,  
Wuhan Institute of Technology, Wuhan, 430073, Peop. Rep. China

SO Journal of Liquid Chromatography & Related Technologies (2008),  
31(17), 2554-2574

CODEN: JLCTFC; ISSN: 1082-6076

PB Taylor & Francis, Inc.

DT Journal

LA English

AB Polymers were synthesized, resp., by the copolymn. of (1S,2R)-(+)-2-  
amino-1,2-diphenylethanol with 1,4-phenylene diisocyanate (I); and  
(1S,2R)-(+)-2-amino-1,2-diphenylethanol with 1,4-phenylene  
diisocyanate and terephthaloyl chloride (II). The corresponding  
chiral stationary phases, CSPs I and II, were prepared by  
immobilizing these polymers on 3-aminopropyl silica gel. The  
enantiosepn. ability of obtained chiral stationary phases was  
evaluated with chiral analytes. The effects of organic additives,  
mobile phase composition, temperature, and substituents of chiral  
analytes on enantiosepn. were studied in HPLC. The preliminary  
studies demonstrated that the enantiosepn. ability could be resumed,  
although the chiral stationary phase experienced acidic mobile phase.

IT 618-36-0, (+)-1-Amino-1-phenylethane 3886-69-9,  
2627-86-3, (-)-1-Amino-1-phenylethane  
(+)-1-Amino-1-phenylethane

RL: ANT (Analyte); ANST (Analytical study)

(preparation and enantiosepn. of polymer-type chiral stationary

phases

derived from chiral aminodiphenylethanol)

RN 618-36-0 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 80-4 (Organic Analytical Chemistry)

IT	50-35-1	77-21-4	125-84-8	602-09-5,		
	(±)-2,2'-Dihydroxy-1,1'-Binaphthalene	618-36-0,				
	(±)-1-Amino-1-phenylethane	744-80-9	841-67-8	2614-06-4		
	2627-86-3, (-)-1-Amino-1-phenylethane	3480-59-9				
	3886-69-9, (+)-1-Amino-1-phenylethane	4108-58-1				
	6452-71-7	17575-58-5	17575-59-6	18006-57-0	18531-94-7,	
	(+)-2,2'-Dihydroxy-1,1'-Binaphthalene	18531-99-2,				
	(-)-2,2'-Dihydroxy-1,1'-Binaphthalene	19035-02-0	20826-48-6			
	22916-47-8	22972-96-9	27220-47-9	27523-40-6	29270-30-2	
	31576-00-8	39562-70-4	47447-52-9	47447-53-0	55511-44-9	
	57288-03-6	66648-29-1	67648-61-7	71283-66-4	73094-37-8	
	73094-39-0	76703-62-3	76703-65-6	80873-62-7	80890-07-9	
	84057-95-4	91465-08-6	94050-90-5	98626-61-0	98717-16-9	
	105118-15-8	109579-04-6	113960-28-4	113960-29-5	125811-10-1	
	155236-70-7	256398-61-5	322764-96-5	322764-97-6	853788-61-1	

928007-57-2 931385-15-8 931385-16-9 931385-18-1 931385-19-2  
 948579-27-9 948579-28-0 948579-29-1 1100200-39-2

RL: ANT (Analyte); ANST (Analytical study)

(preparation and enantiosepn. of polymer-type chiral stationary

phases

derived from chiral aminodiphenylethanol)

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1  
 CITINGS)

RE.CNT 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L39 ANSWER 2 OF 7 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2004:778540 HCAPLUS Full-text

DN 141:279440

TI Phosgenation process for the production of polyisocyanates from  
 primary amines and phosgene

IN Brodhagen, Andreas; Sohn, Martin; Nevejans, Filip; Stroefer,  
 Eckhard; Woelfert, Andreas; Oehlschlaeger, Steffen

PA BASF A.-G., Germany

SO Ger. Offen., 7 pp.

CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-----	----	-----	-----	
PI	DE 10310888	A1	20040923	DE 2003-10310888	200303 11
	WO 2004080587	A1	20040923	WO 2004-EP1673	200402 20
W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW:	BW, BH, BG, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
EP	1601456	A1	20051207	EP 2004-713043	200402



20

EP 1601456 B1 20090930  
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC,  
 PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU,  
 SK  
 CN 1758956 A 20060412 CN 2004-80006608 200402  
 20  
 CN 100374196 C 20080312  
 JP 2006519793 T 20060831 JP 2006-504443 200402  
 20  
 AT 444118 T 20091015 AT 2004-713043 200402  
 20  
 ES 2331184 T3 20091223 ES 2004-713043 200402  
 20  
 MX 2005008907 A 20051005 MX 2005-8907 200508  
 22  
 US 20060223966 A1 20061005 US 2005-546890 200508  
 24

PRAI DE 2003-10310888 A 20030311  
 WO 2004-EP1673 W 20040220

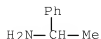
## ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB A procedure is described for the production of polyisocyanates by the phosgenation of primary amines with phosgene, which process comprises: (A) mixture of the amine by the phosgene; (B) conversion of the amine by the phosgene in a retention-time reactor; and, optionally (C) transfer of the reactor output from step (B) into a distillation column. The process is characterized that the retention-time reactor in step (B) is configured as a plug-flow system.

IT 618-36-0, (1-Phenylethyl)amine  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (phosgenation process for the production of polyisocyanates from primary amines and phosgene using)

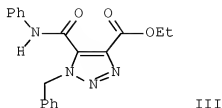
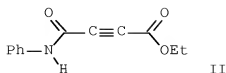
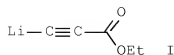
RN 618-36-0 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



IC ICM C07C263-10  
 CC 45-4 (Industrial Organic Chemicals, Leather, Fats, and Waxes)  
 Section cross-reference(s): 48  
 ST polyisocyanate manuf primary amine phosgenation process;  
 diisocyanate manuf primary amine phosgenation process  
 IT 87-62-7 108-00-9 110-58-7, 1-Aminopentane 121-05-1 124-09-4,  
 Hexamethylenediamine, reactions 543-82-8, 2-Amino-6-methylheptane  
 599-61-1, 3,3'-Diaminodiphenyl sulfone 618-36-0,  
 (1-Phenylethyl)amine 1003-03-8, Cyclopentylamine 1572-55-0,  
 4-(Aminomethyl)-1,8-octanediamine 2479-47-2,  
 2,2-Bis(4-aminophenyl)propane 2855-13-2, Isophoronediamine  
 22374-89-6 26764-44-3 38096-30-9, Diaminonaphthalene  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (phosgenation process for the production of polyisocyanates from  
 primary amines and phosgene using)

L39 ANSWER 3 OF 7 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2004:395121 HCAPLUS Full-text  
 DN 141:71493  
 TI 1,3-Dipolar cycloadditions of organic azides to ester- or  
 benzotriazolylcarbonyl-activated acetylenic amides  
 AU Katritzky, Alan R.; Zhang, Yuming; Singh, Sandeep K.; Steel, Peter  
 J.  
 CS Cent. Heterocyclic Compounds, Dep. Chem., Univ. Florida,  
 Gainesville, FL, 32611-7200, USA  
 SO ARKIVOC (Gainesville, FL, United States) (2003), (15), 47-64  
 CODEN: AGFUAR  
 URL: [http://arkat-usa.org/ark/journal/2003/General\\_Part\(xv\)/03-912A/03-912A.pdf](http://arkat-usa.org/ark/journal/2003/General_Part(xv)/03-912A/03-912A.pdf)  
 PB Arkat USA Inc.  
 DT Journal; (online computer file)  
 LA English  
 OS CASREACT 141:71493  
 GI



AB Reactions of 3-lithiopropiolate I with isocyanates or diisocyanates gave mono-carbamoylpropiolates, e.g., II and bis-carbamoylpropiolates. 1,3-Dipolar cycloaddns. of benzyl azide with mono-acetylenes under thermal conditions gave mono-triazoles, e.g., III. The structure of III was confirmed by X-ray crystallog. Microwave induced cycloaddns. of mono-azide with bis-carbamoylpropiolates furnished the corresponding bis-triazoles. Similar reactions of 3-(azidomethyl)-3-methyloxetane with mono-acetylenes or bis-acetylenes produced the mono- and bis-triazoles. Reactions of 1,4-bis(azidomethyl)benzene with mono-acetylenes gave the azido-triazoles and microwave irradiation with simultaneous air-cooling gave bis-triazoles. 1,3-Dipolar cycloaddn. of benzotriazolylcarbonyl-substituted acetylene and benzyl azide proceeded smoothly under microwave irradiation or thermal conditions to give the corresponding triazole, which on further treatment with a variety of amines gave the C-carbamoyl triazoles.

IT 618-36-0,  $\alpha$ -Methylbenzylamine

RL: RCT (Reactant); RACT (Reactant or reagent)  
(regioselective preparation of triazolecarboxamides via heterocyclization of phenylpropynoylbenzotriazole with benzyl azide followed by substitution with amines)

RN 618-36-0 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



CC 28-10 (Heterocyclic Compounds (More Than One Hetero Atom))  
 Section cross-reference(s): 25, 75

IT 618-36-0,  $\alpha$ -Methylbenzylamine  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (regioselective preparation of triazolecarboxamides via  
 heterocyclization of phenylpropynoylbenzotriazole with benzyl  
 azide followed by substitution with amines)

OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3  
 CITINGS)

RE.CNT 63 THERE ARE 63 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L39 ANSWER 4 OF 7 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2003:772353 HCAPLUS Full-text  
 DN 139:395586

TI Highly Cooperative Formation of Bis-Urea Based Supramolecular  
 Polymers

AU Simic, Vesna; Bouteiller, Laurent; Jalabert, Matthieu  
 CS Laboratoire de Chimie des Polymeres UMR 7610, Universite Pierre et  
 Marie Curie, Paris, 75252, Fr.

SO Journal of the American Chemical Society (2003), 125(43),  
 13148-13154  
 CODEN: JACSAT; ISSN: 0002-7863

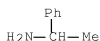
PB American Chemical Society  
 DT Journal  
 LA English  
 OS CASREACT 139:395586

AB Eleven bis-ureas were synthesized, and some of their properties are  
 reported. Several of these compds. form supramol. polymers in  
 organic solvents. The self-association is shown by FTIR spectroscopy  
 to display cooperativity at two levels. The first level of  
 cooperativity is due to the synergistic association of the two urea  
 functions of a single mol. The second level of cooperativity is  
 revealed by the fact that the formation of dimers is less favored  
 than that of long oligomers.

IT 618-36-0  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (highly cooperative formation of bis-urea based supramol.  
 polymers)

RN 618-36-0 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



CC 22-12 (Physical Organic Chemistry)  
 Section cross-reference(s): 68

IT 91-08-7, 2,6-Toluene diisocyanate 104-75-6,  
 2-Ethylhexylamine 110-58-7, Pentylamine 584-84-9, 2,4-Toluene  
 diisocyanate  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (conversion to bis-urea derivative; highly cooperative formation  
 of  
 bis-urea based supramol. polymers)

IT 107-45-9 107-85-7, 3-Methylbutylamine 111-86-4, Octylamine  
 124-30-1, Octadecylamine 543-82-8, 1,5-Dimethylhexylamine  
 614-68-6, 2-Tolylisocyanate 616-24-0, 1-Ethylpropylamine  
 618-36-0 622-58-2, 4-Tolylisocyanate  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (highly cooperative formation of bis-urea based supramol.  
 polymers)

OSC.G 57 THERE ARE 57 CAPLUS RECORDS THAT CITE THIS RECORD (59  
 CITINGS)

RE.CNT 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L39 ANSWER 5 OF 7 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2001:165790 HCAPLUS Full-text

DN 134:207403

TI Improved procedure for the production of mono- and oligoisocyanates  
 by the phosgenation of primary amines in the presence of catalytic  
 amounts of monoisocyanates

IN Stamm, Armin; Kneuper, Heinz-josef; Thil, Lucien; Henkelmann, Jochem

PA BASF AG, Germany

SO Ger. Offen., 8 pp.  
 CODEN: GWXXBX

DT Patent

LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-----	---	-----	-----	
PI	DE 19942299	A1	20010308	DE 1999-19942299	199909 04

WO 2001017951 A1 20010315 WO 2000-EP8221 200008  
23

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH,  
CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH,  
GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK,  
LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ,  
PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ,  
UA, UG, US, UZ, VN, YU, ZA, ZW

RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH,  
CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE,  
BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG

EP 1208082 A1 20020529 EP 2000-951530 200008  
23

EP 1208082 B1 20040407

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC,  
PT, IE, SI, LT, LV, FI, RO, MK, CY, AL

AT 263752 T 20040415 AT 2000-951530 200008  
23

US 6683204 B1 20040127 US 2002-70393 200203  
04

PRAI DE 1999-19942299 A 19990904  
WO 2000-EP8221 W 20000823

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS CASREACT 134:207403; MARPAT 134:207403

AB Aliphatic, cycloaliph., araliph. [e.g., R-(+)-phenylethyl  
isocyanate], or aromatic mono- and oligoisocyanates are prepared in  
high yield and selectivity by the phosgenation of the appropriate  
primary amines [e.g., R-(+)-phenylethylamine] at atmospheric pressure  
with phosgene in the presence of catalytic amts. of monoisocyanates  
(e.g., Bu isocyanate) in an inert solvent (e.g., chlorobenzene).

IT 2627-86-3 3886-69-9

RL: RCT (Reactant); RACT (Reactant or reagent)  
(improved procedure for the production of mono- and  
oligoisocyanates  
by the phosgenation of primary amines in the presence of  
catalytic amts. of monoisocyanates)

RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



IC ICM C07B043-10

ICS C07C263-10

CC 21-2 (General Organic Chemistry)

Section cross-reference(s): 25, 45

IT 62-53-3, Aniline, reactions 75-44-5, Phosgene 108-91-8,  
Aminocyclohexane, reactions 124-09-4, Hexamethylenediamine,  
reactions 2627-86-3 3886-69-9 26764-44-3

RL: RCT (Reactant); RACT (Reactant or reagent)

(improved procedure for the production of mono- and

oligoisocyanates

by the phosgenation of primary amines in the presence of  
catalytic amts. of monoisocyanates)

IT 103-71-9P, Phenyl isocyanate, preparation 2855-13-2P, Isophorone  
diamine 3173-53-3P, Cyclohexyl isocyanate 4098-71-9P, Isophorone  
diisocyanate 33375-06-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(improved procedure for the production of mono- and

oligoisocyanates

by the phosgenation of primary amines in the presence of  
catalytic amts. of monoisocyanates)

L39 ANSWER 6 OF 7 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 1995:995087 HCAPLUS [Full-text](#)

DN 124:97810

OREF 124:18085a,18088a

TI Process and composition for preparing a dental polymer product

IN Klee, Joachim E.; Leube, Walter

PA Dentsply GMBH, Germany

SO Eur. Pat. Appl., 21 pp.

CODEN: EPXXDW

DT Patent

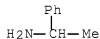
LA English

FAN.CNT 5

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-----	----	-----	-----	
PI	EP 678533	A2	19951025	EP 1995-105945	199504 20
	EP 678533	A3	19980128		
	EP 678533	B1	20040901		
	R: BE, CH, DE, DK, ES, FR, GB, IT, LI, NL, SE				
	CA 2146816	A1	19951023	CA 1995-2146816	199504 11
	NO 9501494	A	19951023	NO 1995-1494	199504 20
	NO 309610	B1	20010226		
	FI 9501909	A	19951023	FI 1995-1909	199504 21
	FI 116294	B1	20051031		
	ZA 9503252	A	19960219	ZA 1995-3252	199504 21
	US 6369164	B1	20020409	US 1996-582235	199601 03
	US 5876210	A	19990302	US 1996-754664	199611 21
	US 20020143108	A1	20021003	US 2002-54360	200201 22
	US 20050043490	A1	20050224	US 2004-938459	200409 10
PRAI	US 1994-231535	A	19940422		
	US 1993-67774	B2	19930526		
	US 1994-217998	A2	19940325		
	US 1994-359217	B1	19941219		
	US 1996-582235	A1	19960103		
	US 2002-54360	B1	20020122		



- AB The invention concerns a process for preparing a polymer composition, that is free-radical/photochem. and thermal curing of epoxide-methacrylate and/or isocyanate-methacrylate adhesives in broadest terms, dental/medical adhesives, and dental restoratives. Furthermore the dual curing of epoxide-methacrylate and/or isocyanate-methacrylate adhesives can be used in the optical industry, in optoelectronics and microelectronics, for example for the adhesion of complicated optical components in the combination glass/glass, glass/metal. Advantageous is the small shrinkage during polymerization and the good mech. properties in combination with the possibility of step-wise or one-step polymerization
- IT 618-36-0,  $\alpha$ -Phenethylamine  
 RL: POF (Polymer in formulation); THU (Therapeutic use); BIOL (Biological study); USES (Uses)  
 (macromonomers and polymer compns. for dental applications)
- RN 618-36-0 HCAPLUS
- CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



- IC ICM C08F290-06  
 ICS A61K006-08
- CC 63-7 (Pharmaceuticals)  
 Section cross-reference(s): 35, 36
- IT 56-18-8, Dipropylenetriamine 78-67-1, Azobisisobutyronitrile  
 79-41-4, biological studies 80-05-7, biological studies 80-09-1  
 85-42-7, Hexahydrophthalic acid anhydride 85-43-8,  
 Tetrahydrophthalic acid anhydride 85-44-9, 1,3-Isobenzofurandione  
 88-99-3, 1,2-Benzenedicarboxylic acid, biological studies 94-25-7,  
 p-Aminobenzoic acid butyl ester 94-36-0, Dibenzoylperoxide,  
 biological studies 100-21-0, 1,4-Benzenedicarboxylic acid,  
 biological studies 100-46-9, Benzylamine, biological studies  
 101-68-8 106-91-2 106-91-2D, reaction product  
 withbisphenolAdiglycidyl and dibenzyl-5-oxanonane-1,9-diamine  
 108-30-5, biological studies 108-46-3, 1,3-Benzenediol, biological  
 studies 108-80-5, Cyanuric acid 108-95-2, Phenol, biological  
 studies 109-16-0 110-15-6, Butanedioic acid, biological studies  
 110-70-3, N,N'-Dimethylethylenediamine 123-31-9, 1,4-Benzenediol,  
 biological studies 124-04-9, Hexanedioic acid, biological studies  
 128-37-0, BHT, biological studies 140-28-3, N,N'  
 Dibenzylethylenediamine 141-43-5, biological studies 378-46-1

618-36-0,  $\alpha$ -Phenethylamine 768-94-5,  
 1-Adamantanamine 822-06-0 1122-17-4, Dichloromaleic acid  
 anhydride 1321-14-8, Thiocol 1565-94-2D, Bis-GMA, ethoxylated  
 1675-54-3 1687-30-5, Hexahydrophthalic acid 2095-03-6, Bisphenol  
 F diglycidyl ether 2358-84-1, Diethylene glycol bismethacrylate  
 2855-13-2 3077-12-1, N,N-Bis( $\beta$ -hydroxyethyl)-p-toluidine  
 3236-53-1 3236-54-2 3524-62-7, Benzoin methyl ether 4098-71-9,  
 Isophorone diisocyanate 4100-80-5 4605-14-5,  
 Tripropylenetetramine 7664-38-2, Phosphoric acid, biological  
 studies 9011-05-6, Urea-formaldehyde resin 9046-10-0  
 10193-95-0 10373-78-1, Camphor quinone 13598-36-2, Phosphonic  
 acid 14970-87-7, 1,8-Dimercapto-3,6-dioxaoctane 15716-30-0  
 16128-67-9 16938-22-0 21544-03-6 25085-99-8 26471-62-5,  
 Toluenediisocyanate 28768-32-3 42450-83-9 66582-26-1,  
 N,N'-Dibenzyl-3,6-dioxaoctane-1,8-diamine 76364-76-6 77125-27-0  
 77125-28-1 113506-22-2, N,N'-Dibenzyl-5-oxanonane-1,9-diamine  
 113506-23-3D, reaction product with 2,3-epoxypropyl methacrylate  
 144450-30-6 172779-90-7

RL: POF (Polymer in formulation); THU (Therapeutic use); BIOL  
 (Biological study); USES (Uses)  
 (macromonomers and polymer compns. for dental applications)

OSC.G 7 THERE ARE 7 CAPLUS RECORDS THAT CITE THIS RECORD (7  
 CITINGS)

L39 ANSWER 7 OF 7 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 1993:233487 HCAPLUS Full-text

DN 118:233487

OREF 118:40423a,40426a

TI Preparation of oxime carbamates from urea, alkanone oximes and  
 alkanamines

IN Leung, Tak W.; Best, Donald C.; Dombek, Bernard D.

PA Union Carbide Chemicals and Plastics Technology Corp., USA

SO U.S., 10 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

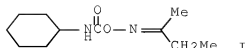
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	US 5179223	A	19930112	US 1990-627196	199012 13

PRAI US 1990-627196 19901213

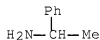
ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS CASREACT 118:233487; MARPAT 118:233487

GI



- AB A process for producing oxime carbamates is claimed which comprises contacting a mixture of primary amine, a urea and at least one oxime component. Said oxime carbamates can be thermally decomposed to give isocyanates. A mixture of cyclohexanamine (10 g), urea (6.1 g) and 2-butanone oxime (130 g) was refluxed at 150° for 4 h to give (1-methylpropylidene)amino N-cyclohexylcarbamate [O-[(cyclohexylamino)carbonyl]-2-butanone oxime] (I) in 87% yield. A mixture of Jeffamine D-2000, urea and 2-butanone oxime gave an oxime carbamate derivative having an IR absorption at 1730 cm<sup>-1</sup>. This oxime carbamate was fed into the hot finger of a falling film evaporator (elimination of 2-butanone oxime) to give a diisocyanate derivative with an isocyanate content of 2.5%.
- IT 618-36-0, α-Methylbenzylamine  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (reaction of, with urea and alkanone oxime, alkylidenamino  
 N-alkylcarbamate from)
- RN 618-36-0 HCAPLUS
- CN Benzenemethanamine, α-methyl- (CA INDEX NAME)



- IC ICM C07C269-04
- INCL 560033000
- CC 23-16 (Aliphatic Compounds)  
 Section cross-reference(s): 35
- IT 106-49-0, (4-Methylphenyl)amine, reactions 107-11-9, Allylamine  
 108-91-8, Cyclohexanamine, reactions 618-36-0,  
 α-Methylbenzylamine 2855-13-2, Isophorone diamine  
 RL: RCT (Reactant); RACT (Reactant or reagent)

(reaction of, with urea and alkanone oxime, alkylidenamino  
N-alkylcarbamate from)

RE.CNT 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> d 129 1-12 bib abs hitstr hitind

L29 ANSWER 1 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
AN 2008:223135 HCAPLUS Full-text  
DN 148:402890  
TI Structural aspects of nucleation inhibitors for diastereomeric  
resolutions and the relationship to Dutch Resolution  
AU Leeman, Michel; Brasile, Giuseppina; Gelens, Edith; Vries, Ton;  
Kaptein, Bernard; Kellogg, Richard  
CS Syncom BV, Groningen, 9747, Neth.  
SO Angewandte Chemie, International Edition (2008), 47(7), 1287-1290  
CODEN: ACIEF5; ISSN: 1433-7851  
PB Wiley-VCH Verlag GmbH & Co. KGaA  
DT Journal  
LA English  
OS CASREACT 148:402890  
AB Nucleation inhibitors for use in the Dutch resolution of  
diastereomers of racemic 3-methoxyphenylethylamine by selective  
crystallization with (S)- or (R)-mandelic acid have been designed and  
tested.  
IT 2627-86-3, (S)-1-Phenylethyl amine 3886-69-9  
RL: MOA (Modifier or additive use); USES (Uses)  
(kinetic resolution of diastereomers of methoxyphenylethylamine  
via Dutch resolution with chiral mandelic acid and various nucleation  
inhibitors)  
RN 2627-86-3 HCAPLUS  
CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS  
CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 25-9 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds)  
 IT 76-93-7, uses 77-92-9, Citric acid, uses 79-14-1, Glycolic acid,  
 uses 79-33-4, uses 92-70-6 144-62-7, Oxalic acid, uses  
 450-52-2 492-37-5 492-86-4, 4-Chloromandelic acid 515-30-0  
 585-32-0 827-97-4 1477-55-0, 1,3-Benzenedimethanamine  
 2627-86-3, (S)-1-Phenylethyl amine 2743-38-6  
 3886-69-9 6064-63-7 6298-96-0, 1-(4-Methoxyphenyl)ethyl  
 amine 6940-50-7, 4-Bromomandelic acid 7322-88-5 7326-19-4  
 10421-85-9, 2-Chloromandelic acid 20445-31-2 26164-26-1  
 29841-69-8 31284-89-6 46065-10-5 49839-81-8 65148-70-1,  
 3-Methylmandelic acid 68969-02-8 70138-19-1,  
 1-(3-Methylphenyl)ethyl amine 71707-27-2, 4-Benzyloxymandelic acid  
 698378-52-8 870196-09-1

RL: MOA (Modifier or additive use); USES (Uses)

(kinetic resolution of diastereomers of methoxyphenylethylamine

via

Dutch resolution with chiral mandelic acid and various nucleation  
 inhibitors)

OSC.G 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2  
 CITINGS)

RE.CNT 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 2 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2007:1300933 HCAPLUS Full-text

DN 147:525037

TI C12-20-Fatty acid salts with amines, alkanolamines, and alkali  
 metals as antistain additives for aqueous metalworking oils

IN Brutto, Patrick E.; Pyzowski, Bonnie A.; Coburn, Charles E.

PA Angus Chemical Company, USA

SO PCT Int. Appl., 29 pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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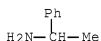
PI	WO 2007130836	A1	20071115	WO 2007-US67462	200704 26
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, MT, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
EP	2027237	A1	20090225	EP 2007-761320	200704 26
	R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, LV, MC, MT, NL, PL, PT, RO, SE, SI, SK, TR, AL, BA, HR, MK, RS				
JP	2009536254	T	20091008	JP 2009-509962	200704 26
US	20090170736	A1	20090702	US 2008-297675	200810 20
CN	101437929	A	20090520	CN 2007-80015489	200810 28
IN	2008CN06007	A	20090403	IN 2008-CN6007	200811 05
KR	2009018940	A	20090224	KR 2008-729871	200812 05
PRAI	US 2006-746549P	P	20060505		
	WO 2007-US67462	W	20070426		

# ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB Antistain additives for aqueous metalworking fluids (with pH  $\geq 7$ ) consist of C12-20-linear and branched fatty acids neutralized with  $\geq 1$  of an amine, alkanolamine, and an alkali metal hydroxide. The neutralized fatty acids are present in  $\geq 0.10$  weight% concentration in the finished metalworking oil, and  $\geq 1$  weight% in the metalworking oil concentrate. The stain inhibitors are especially useful for

metalworking of nonferrous alloys, especially Al alloys, and ferrous alloys (e.g., galvanized steel).

- IT 618-36-0D,  $\alpha$ -Phenylethylamine, compds. with C12-20-linear and branched fatty acids  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (antistain additives; C12-20-fatty acid salts with amines, alkanolamines, and alkali metals as antistain additives for aqueous metalworking oils)
- RN 618-36-0 HCAPLUS
- CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



- CC 51-8 (Fossil Fuels, Derivatives, and Related Products)  
 Section cross-reference(s): 56
- IT 57-10-3D, Palmitic acid, salts with amines, alkanolamines, and alkali metal hydroxides 57-11-4D, Stearic acid, salts with amines, alkanolamines, and alkali metal hydroxides 60-33-3D, Linoleic acid, salts with amines, alkanolamines, and alkali metal hydroxides 62-53-3D, Aniline, compds. with C12-20-linear and branched fatty acids 64-04-0D,  $\beta$ -Phenylethylamine, compds. with C12-20-linear and branched fatty acids 74-89-5D, Methylamine, compds. with C12-20-linear and branched fatty acids 75-04-7D, Ethylamine, compds. with C12-20-linear and branched fatty acids 75-31-0D, Isopropylamine, compds. with C12-20-linear and branched fatty acids 75-50-3D, Trimethylamine, compds. with C12-20-linear and branched fatty acids 75-59-2D, Tetramethylammonium hydroxide, compds. with C12-20-linear and branched fatty acids 75-64-9D, tert-Butylamine, compds. with C12-20-linear and branched fatty acids 78-81-9D, Isobutylamine, compds. with C12-20-linear and branched fatty acids 78-96-6D, Monoisopropanolamine, compds. with tall-oil fatty acids 90-04-0D, o-Anisidine, compds. with C12-20-linear and branched fatty acids 92-87-5D, Benzidine, compds. with C12-20-linear and branched fatty acids 95-51-2D, o-Chloroaniline, compds. with C12-20-linear and branched fatty acids 95-53-4D, o-Toluidine, compds. with C12-20-linear and branched fatty acids 100-46-9D, Benzylamine, compds. with C12-20-linear and branched fatty acids 100-61-8D, N-Methylaniline, compds. with C12-20-linear and branched fatty acids 101-83-7D, Dicyclohexylamine, compds. with C12-20-linear and branched fatty acids 102-69-2D,

Tri-n-propylamine, compds. with C12-20-linear and branched fatty acids 102-71-6D, Triethanolamine, compds. with C12-20-linear and branched fatty acids 104-94-9D, p-Anisidine, compds. with C12-20-linear and branched fatty acids 106-47-8D, p-Chloroaniline, compds. with C12-20-linear and branched fatty acids 106-49-0D, p-Toluidine, compds. with C12-20-linear and branched fatty acids 107-10-8D, n-Propylamine, compds. with C12-20-linear and branched fatty acids 107-15-3D, Ethylenediamine, compds. with C12-20-linear and branched fatty acids 108-42-9D, m-Chloroaniline, compds. with C12-20-linear and branched fatty acids 108-44-1D, m-Toluidine, compds. with C12-20-linear and branched fatty acids 108-91-8D, Cyclohexylamine, compds. with C12-20-linear and branched fatty acids 109-73-9D, n-Butylamine, compds. with C12-20-linear and branched fatty acids 109-89-7D, Diethyl amine, compds. with C12-20-linear and branched fatty acids 110-60-1D, Tetramethylenediamine, compds. with C12-20-linear and branched fatty acids 110-97-4D, Diisopropanolamine, compds. with C12-20-linear and branched fatty acids 111-42-2D, Diethanolamine, compds. with C12-20-linear and branched fatty acids 111-75-1D, n-Butylethanolamine, compds. with tall-oil fatty acids 112-80-1D, Oleic acid, salts with amines, alkanolamines, and alkali metal hydroxides 115-70-8D, 2-Amino-2-ethyl-1,3-propanediol, compds. with C12-20-linear and branched fatty acids 121-44-8D, Triethylamine, compds. with C12-20-linear and branched fatty acids 122-20-3D, Triisopropanolamine, compds. with C12-20-linear and branched fatty acids 124-09-4D, Hexamethylenediamine, compds. with C12-20-linear and branched fatty acids 124-40-3D, Dimethylamine, compds. with C12-20-linear and branched fatty acids 124-68-5D, compds. with tall-oil fatty acids 141-22-0D, Ricinoleic acid, salts with amines, alkanolamines, and alkali metal hydroxides 141-43-5D, Monoethanolamine, compds. tall-oil fatty acids 142-84-7D, Di-n-propylamine, compds. with C12-20-linear and branched fatty acids 143-07-7D, Lauric acid, salts with amines, alkanolamines, and alkali metal hydroxides 506-32-1D, Arachidonic acid, salts with amines, alkanolamines, and alkali metal hydroxides 536-90-3D, m-Anisidine, compds. with C12-20-linear and branched fatty acids 544-63-8D, Myristic acid, salts with amines, alkanolamines, and alkali metal hydroxides 618-36-0D,  $\alpha$ -Phenylethylamine, compds. with C12-20-linear and branched fatty acids 929-06-6D, Diglycolamine, compds. with tall-oil fatty acids 1189-37-3D, Pristanic acid, salts with amines, alkanolamines, and alkali metal hydroxides 7664-41-7D, Ammonia, salts with C12-20-linear and branched fatty acids 10339-73-8D, 4,8,12-Trimethyltridecanoic acid, salts with amines, alkanolamines, and alkali metal hydroxides 13952-84-6D, sec-Butylamine, compds. with C12-20-linear and branched fatty acids 14721-66-5D, Phytanic acid, salts with amines, alkanolamines, and alkali metal hydroxides



23247-33-8 25354-97-6D, 2-Hexyldecanoic acid, salts with amines, alkanolamines, and alkali metal hydroxides 29106-32-9D, Chaulmoogric acid, salts with amines, alkanolamines, and alkali metal hydroxides 30399-84-9D, Isostearic acid, salts with amines, alkanolamines, and alkali metal hydroxides 50862-89-0 56669-89-7 68140-41-0 68239-05-4 68815-69-0 93920-23-1 93981-99-8 125111-39-9D, 2-Cyclohexene-1-octanoic acid, salts with amines, alkanolamines, and alkali metal hydroxides 404875-53-2 929700-37-8 956595-73-6D, salts with amines, alkanolamines, and alkali metal hydroxides 956595-79-2 956595-80-5  
 RL: MOA (Modifier or additive use); USES (Uses)

(antistain additives; C12-20-fatty acid salts with amines, alkanolamines, and alkali metals as antistain additives for

aqueous

metalworking oils)

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 3 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2006:1186424 HCAPLUS Full-text

DN 146:101538

TI Induction of chirality into a fully sulfonated poly(methoxyaniline) via acid-base interactions with chiral amines

AU Strounina, Ekaterina V.; Kane-Maguire, Leon A. P.; Wallace, Gordon G.

CS ARC Centre of Excellence for Electromaterials Science, Intelligent Polymer Research Institute, University of Wollongong, Wollongong, 2522, Australia

SO Polymer (2006), 47(24), 8088-8094  
 CODEN: POLMAG; ISSN: 0032-3861

PB Elsevier Ltd.

DT Journal

LA English

AB A wide range of chiral amines and amino alcs. associate with poly(2-methoxyaniline-5-sulfonic acid) (PMAS) in aqueous solution, from which optically active PMAS·(amine) films can be cast. The chiral induction is believed to be initiated by acid-base interactions with "free" sulfonic acid groups on the PMAS chains. Chiral amine:PMAS dimer molar ratios as low as 1:4 give PMAS·(amine) films with similar optical activity to those cast from 1:1 M mixts., indicating that only one in four of the "free" sulfonate groups on the PMAS chains need to be electrostatically bound by chiral ammonium ions to achieve optimal chiral induction. CD studies show that the enantiomeric amines (R)-(+)- and (S)-(-)-1-phenylethylamine induce the opposite helical hands for the supermol. assemblies of PMAS chains. However, there is no clear correlation between the sign of the CD signals for

the PMAS·(amine) films and the configuration of structurally diverse amines.

IT 2627-86-3, (S)-(-)-1-Phenylethylamine 3886-69-9  
 , (R)-(+)-1-Phenylethylamine  
 RL: MOA (Modifier or additive use); PRP (Properties); USES  
 (Uses)  
 (induction of chirality into fully sulfonated  
 poly(methoxyaniline) via acid-base interactions with chiral  
 amines)

RN 2627-86-3 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 37-5 (Plastics Manufacture and Processing)

Section cross-reference(s): 36

IT 2627-86-3, (S)-(-)-1-Phenylethylamine 3886-69-9  
 , (R)-(+)-1-Phenylethylamine 17430-98-7 18531-95-8 19131-99-8  
 23364-44-5, (1S,2R)-(+)-2-Amino-1,2-diphenylethanol 35320-23-1,  
 (R)-(-)-2-Amino-1-propanol

RL: MOA (Modifier or additive use); PRP (Properties); USES  
 (Uses)

(induction of chirality into fully sulfonated  
 poly(methoxyaniline) via acid-base interactions with chiral  
 amines)

OSC.G 6 THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (6  
 CITINGS)

RE.CNT 27 THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD

## ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 4 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2004:1151660 HCAPLUS Full-text  
 DN 142:412190  
 TI Stabilization of single-wall carbon nanotubes in fully sulfonated polyaniline  
 AU Panhuis, Marc in het; Kane-Maguire, Leon A. P.; Moulton, Simon E.; Innis, Peter C.; Wallace, Gordon G.  
 CS ARC Centre for Nanostructured Electromaterials, Intelligent Polymer Research Institute, University of Wollongong, NSW 2522, Australia  
 SO Journal of Nanoscience and Nanotechnology (2004), 4(8), 976-981  
 CODEN: JNNOAR; ISSN: 1533-4880  
 PB American Scientific Publishers  
 DT Journal  
 LA English  
 AB The interaction of single wall carbon nanotubes (SWNT) with an aqueous solution of the fully sulfonated polyaniline poly(2-methoxyaniline-5-sulfonic acid) (PMAS) and (+)-1-phenylethylamine (PhEA) has been investigated using spectroscopic methods. UV-vis spectral measurements show that the PMAS backbone undergoes conformational changes upon interaction with both SWNT and PhEA. Partial intercalation of PMAS into SWNT bundles was confirmed by Raman spectroscopy and electron microscopy.  
 IT 3886-69-9, (+)-1-Phenylethylamine  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (stabilization of single-wall carbon nanotubes in fully sulfonated polyaniline)  
 RN 3886-69-9 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 37-6 (Plastics Manufacture and Processing)  
 IT 3886-69-9, (+)-1-Phenylethylamine  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (stabilization of single-wall carbon nanotubes in fully sulfonated polyaniline)  
 OSC.G 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2 CITINGS)

RE.CNT 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 5 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
AN 2004:831399 HCAPLUS Full-text  
DN 142:6594  
TI Asymmetric catalysis in a micro reactor-Ce, Yb and Lu catalyzed  
enantioselective addition of trimethylsilyl cyanide to benzaldehyde  
AU Joansson, Christina; Lundgren, Stina; Haswell, Stephen J.; Moberg,  
Christina  
CS Department of Chemistry, Organic Chemistry, Royal Institute of  
Technology, Stockholm, SE-100 44, Swed.  
SO Tetrahedron (2004), 60(46), 10515-10520  
CODEN: TETRA; ISSN: 0040-4020  
PB Elsevier B.V.  
DT Journal  
LA English  
OS CASREACT 142:6594  
AB A T-shaped micro reactor was used for the optimization of reaction  
conditions for the enantioselective silylcyanation of benzaldehyde  
catalyzed by lanthanide-pybox complexes. Compared to a conventional  
batch procedure, higher conversion was observed within shorter  
reaction time. The micro reactor process involving Lu(III) afforded  
essentially the same enantioselectivity as the batch process (73 vs.  
76% ee), whereas the enantioselectivity was lower in the micro  
reactor for catalysts containing Yb(III) (53 compared to 72%).  
Ce(III) provided very low selectivity in both types of processes (1  
and 11% ee, resp.). A study of the effect of additives showed that  
the enantioselectivity in the Yb catalyzed reaction performed in the  
micro reactor could be increased to 66%, whereas only a minor  
improvement, to 78% ee, was observed in the reaction with Lu.  
IT 2627-86-3 3886-69-9  
RL: MOA (Modifier or additive use); USES (Uses)  
(influence of additives on enantioselectivity of  
lanthanide-catalyzed addition of trimethylsilyl cyanide to  
benzaldehyde using micro reactor)  
RN 2627-86-3 HCAPLUS  
CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 29-6 (Organometallic and Organometalloidal Compounds)  
 Section cross-reference(s): 22  
 IT 60-29-7, Diethyl ether, uses 64-17-5, Ethanol, uses 67-56-1,  
 Methanol, uses 75-65-0, tert-Butanol, uses 90-39-1, Sparteine  
 109-99-9, Tetrahydrofuran, uses 694-59-7, Pyridine N-oxide  
 791-28-6, Triphenylphosphine oxide 874-52-2, N,N-Dimethylaniline  
 N-oxide 2216-51-5 2627-86-3 3623-51-6, Neomenthol  
 3886-69-9 5824-40-8, Tritylamine 7732-18-5, Water, uses  
 10311-08-7, Dimethylphenylphosphine oxide 14898-79-4,  
 (R)-2-Butanol 15356-60-2, D-Menthol  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (influence of additives on enantioselectivity of  
 lanthanide-catalyzed addition of trimethylsilyl cyanide to  
 benzaldehyde using micro reactor)  
 OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3  
 CITINGS)  
 RE.CNT 46 THERE ARE 46 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT  
 L29 ANSWER 6 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2004:520432 HCAPLUS Full-text  
 DN 141:226117  
 TI Helicity Induction on Poly(phenylacetylene)s Bearing Phosphonic Acid  
 Pendants with Chiral Amines and Memory of the Macromolecular  
 Helicity Assisted by Interaction with Achiral Amines in Dimethyl  
 Sulfoxide  
 AU Onouchi, Hisanari; Kashiwagi, Daisuke; Hayashi, Kiichiro; Maeda,  
 Katsuhiko; Yashima, Eiji  
 CS Department of Molecular Design and Engineering Graduate School of  
 Engineering, Nagoya University, Nagoya, 464-8603, Japan  
 SO Macromolecules (2004), 37(15), 5495-5503  
 CODEN: MAMOBX; ISSN: 0024-9297  
 PB American Chemical Society  
 DT Journal

LA English

AB Two novel stereoregular poly(phenylacetylene)s bearing a phosphonic acid residue (poly-1) and its monoethyl ester (poly-2) as pendants were prepared by the polymerization of di-Et (4-ethynylphenyl)phosphonate followed by hydrolysis of the di-Et ester groups and polymerization of Et (4-ethynylphenyl)phosphonate, resp. The polymers were found to form a predominantly one-handed helical conformation upon complexation with various chiral amines through noncovalent acid-base interactions in DMSO. The complexes exhibited an induced CD (ICD) in the UV-visible region of the polymer backbones. In particular, poly-2 is an induced helical polymer more sensitive to the chirality of amines than poly-1 and poly((4-carboxyphenyl)acetylene) and yields the same Cotton effect sign when complexed with chiral amines of the same absolute configuration. Moreover, the macromol. helicity of poly-1 and poly-2 induced by chiral amines was "memorized" after the chiral amines were completely removed and replaced with various achiral diamines and oligoamines in DMSO. In sharp contrast to the same memory effect on the induced helical poly((4-carboxyphenyl)acetylene), the helical structures of poly-1 and poly-2 could not be efficiently maintained by achiral monoamines. The effect of the structure of the achiral diamines and oligoamines on the efficiency of the helicity retention and the stability of the memorized polymers were also studied.

IT 3886-69-9

RL: MOA (Modifier or additive use); USES (Uses)

(chiral amine; helicity induction on poly(phenylacetylene)s bearing phosphonic acid pendants with chiral amines and memory of macromol. helicity assisted by interaction with achiral amines in DMSO)

RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 36-7 (Physical Properties of Synthetic High Polymers)

IT 513-49-5 3182-95-4 3886-69-9 3886-70-2 5913-13-3  
7480-35-5 7533-40-6 10420-89-0 35320-23-1 56613-80-0

RL: MOA (Modifier or additive use); USES (Uses)

(chiral amine; helicity induction on poly(phenylacetylene)s bearing phosphonic acid pendants with chiral amines and memory of

macromol. helicity assisted by interaction with achiral amines in DMSO)

OSC.G 40 THERE ARE 40 CAPLUS RECORDS THAT CITE THIS RECORD (41 CITINGS)  
 RE.CNT 64 THERE ARE 64 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 7 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2004:220271 HCAPLUS Full-text  
 DN 140:254069  
 TI Chemically modified, natural cork and its use as a support in reactions on solid phase  
 IN Bardaji Rodriguez, Eduard; Albasa Galtes, Gemma  
 PA Surochem, S.L., Spain  
 SO PCT Int. Appl., 21 pp.  
 CODEN: PIXXD2  
 DT Patent  
 LA English  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI	WO 2004022292	A1	20040318	WO 2003-EP9211	20030820
W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
ES	2212729	A1	20040716	ES 2002-2035	20020906
ES	2212729	B1	20051016		
AU	2003266295	A1	20040329	AU 2003-266295	20030820
EP	1554094	A1	20050720	EP 2003-793739	20030820

EP 1554094 B1 20060927  
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC,  
 PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU,  
 SK  
 AT 340681 T 20061015 AT 2003-793739 200308  
 20  
 ES 2270159 T3 20070401 ES 2003-793739 200308  
 20  
 PRAI ES 2002-2035 A 20020906  
 WO 2003-EP9211 W 20030820  
 OS MARPAT 140:254069  
 AB The present invention relates to chemical modified, natural cork so  
 that it contains reactive chemical groups anchored to its surface,  
 preferably amine groups. It also discloses a process to produce the  
 chemical modified, natural cork, as well as its use as a solid  
 support to be applied to both chemical and biochem. processes on  
 solid phase. Thus, modifying a NaOH-pretreated cork with  
 ethylenediamine gave a surface functionalized cork which was used a  
 scavenger in acylation reaction of benzylamine with benzoyl chloride  
 to remove the remaining benzoyl chloride after the reaction.  
 IT 618-36-0, 1-Phenylethylamine 2627-86-3,  
 (-)-1-Phenylethylamine  
 RL: MOA (Modifier or additive use); RCT (Reactant); RACT  
 (Reactant or reagent); USES (Uses)  
 (modifier; chemical modified, natural cork and its use as support  
 in reactions on solid phase)  
 RN 618-36-0 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)

Ph

H2N-CH-Me

RN 2627-86-3 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).





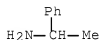
IC ICM B27K007-00  
 CC 35-3 (Chemistry of Synthetic High Polymers)  
 Section cross-reference(s): 43  
 IT 84-95-7, N-Naphthylethylenediamine 100-63-0, Phenylhydrazine  
 107-15-3, Ethylenediamine, reactions 107-35-7, Taurine  
 109-76-2D, Trimethylenediamine, aminopropyl-terminated 124-09-4,  
 Hexamethylenediamine, reactions 143-23-7, Bis(6-aminohexyl)amine  
 618-36-0, 1-Phenylethylamine 623-33-6, Glycine ethyl ester  
 hydrochloride 2627-86-3, (-)-1-Phenylethylamine  
 106392-12-5D, Ethylene oxide-propylene oxide block copolymer,  
 aminopropyl-terminated  
 RL: MOA (Modifier or additive use); RCT (Reactant); RACT  
 (Reactant or reagent); USES (Uses)  
 (modifier; chemical modified, natural cork and its use as support

in reactions on solid phase)  
 RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 8 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2002:831308 HCAPLUS Full-text  
 DN 138:287359  
 TI Simple procedure for preparation of  $\alpha$ -fluoro esters by  
 fluorination of ester enol silyl ethers with perchloryl fluoride  
 AU Fujisawa, Hidehito; Takeuchi, Yoshio  
 CS Faculty of Pharmaceutical Sciences, Toyama Medical and  
 Pharmaceutical University, Toyama, 930-0194, Japan  
 SO Journal of Fluorine Chemistry (2002), 117(2), 173-176  
 CODEN: JFLCAR; ISSN: 0022-1139  
 PB Elsevier Science B.V.  
 DT Journal  
 LA English  
 OS CASREACT 138:287359  
 AB A practical method for general preparation of  $\alpha$ -fluoro esters by  
 fluorination of the corresponding enol silyl ethers using diluted  
 FClO<sub>3</sub> in the presence of suitable amount of amine is described.  
 Fluorination of ester enol silyl ethers in THF at room temperature  
 using diluted perchloryl fluoride (FClO<sub>3</sub>) in the presence of ca. 0.5  
 M eq. of t-BuNH<sub>2</sub> as an additive produced the corresponding  $\alpha$ -fluoro  
 esters in over 80% yields. For example, fluorination of [(1-ethoxy-

2-phenylethenyl)oxy]trimethylsilane with perchloryl fluoride gave  $\alpha$ -fluorobenzenecetic acid Et ester.

IT 618-36-0,  $\alpha$ -Methylbenzenemethanamine  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (simple procedure for preparation of  $\alpha$ -fluoro esters by  
 fluorination of ester enol silyl ethers with perchloryl fluoride)  
 RN 618-36-0 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl- (CA INDEX NAME)



CC 25-18 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds)  
 Section cross-reference(s): 23, 24  
 IT 75-64-9, 2-Methyl-2-propanamine, uses 92-87-5,  
 [1,1'-Biphenyl]-4,4'-diamine 100-61-8, N-Methylbenzenamine, uses  
 108-18-9, N-(1-Methylethyl)-2-propanamine 110-86-1, Pyridine, uses  
 121-44-8, N,N-Diethylethanamine, uses 497-19-8, Carbonic acid  
 disodium salt, uses 584-08-7, Carbonic acid dipotassium salt  
 618-36-0,  $\alpha$ -Methylbenzenemethanamine 7681-49-4,  
 Sodium fluoride (NaF), uses 7789-23-3, Potassium fluoride (KF)  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (simple procedure for preparation of  $\alpha$ -fluoro esters by  
 fluorination of ester enol silyl ethers with perchloryl fluoride)  
 OSC.G 8 THERE ARE 8 CAPLUS RECORDS THAT CITE THIS RECORD (8  
 CITINGS)  
 RE.CNT 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 9 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 2002:71236 HCAPLUS Full-text  
 DN 136:355557  
 TI Novel solid-state polycondensation I. Oxidative-coupling  
 polymerization of 2,6-dihydroxynaphthalene  
 AU Suzuki, Masato; Yatsugi, Yutaka  
 CS Department of Organic and Polymeric Materials, Graduate School of  
 Science and Engineering, and International Research Center of  
 Macromolecular Science, Tokyo Institute of Technology, Meguro-ku,  
 Tokyo, 152-8552, Japan  
 SO Chemical Communications (Cambridge, United Kingdom) (2002), (2),  
 162-163

CODEN: CHCOFS; ISSN: 1359-7345

PB Royal Society of Chemistry

DT Journal

LA English

AB Grinding crystals of 2,6-dihydroxynaphthalene-benzylamine complex with  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  powder in a mortar resulted in the 1,5-oxidative-coupling polymerization of 2,6-dihydroxynaphthalene at room temperature

IT 3886-69-9

RL: MOA (Modifier or additive use); USES (Uses)

(amine derivs. effect on solid state polymerization of 2,6-dihydroxynaphthalene)

RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 35-7 (Chemistry of Synthetic High Polymers)

IT 100-46-9, Benzylamine, uses 107-15-3, Ethylenediamine, uses 539-48-0, p-Xylylenediamine 694-83-7, 1,2-Diaminocyclohexane 1121-22-8 1477-55-0, m-Xylylenediamine 3886-69-9

RL: MOA (Modifier or additive use); USES (Uses)

(amine derivs. effect on solid state polymerization of 2,6-dihydroxynaphthalene)

OSC.G 5 THERE ARE 5 CAPLUS RECORDS THAT CITE THIS RECORD (5 CITINGS)

RE.CNT 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 10 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 2001:599162 HCAPLUS Full-text

DN 136:158

TI Anti-inflammatory planar chiral [2.2]paracyclophaneacetic acid enantiomers

AU Imming, P.; Graf, M.; Tries, S.; Hirschelmann, R.; Krause, E.; Pawlitzki, G.

CS Institut für Pharmazeutische Chemie der Philipps-Universität, Marburg, 35032, Germany

SO Inflammation Research (2001), 50(7), 371-374

CODEN: INREFB; ISSN: 1023-3830

PB Birkhaeuser Verlag  
 DT Journal  
 LA English  
 AB Objective and Design: To elucidate if the planar chiral paracyclophane moiety conveys pharmacol. activity to arylacetic acid analogs in two animal models. Material or Subjects: Female NMRI mice (6 mice/group); female Wistar rats (8 rats/group); thrombocytes from human blood. Treatment: The enantiomers of [2.2]paracyclophaneacetic acid were applied locally (10-7 and 10-6 mol/car) and orally (10-100 mg/kg). Methods: (a) Phorbol myristyl acetate model of acute inflammation of the inner auricle. (b) Oxazolone model of allergic contact dermatitis. (c) Carrageenan model of acute inflammation. (d) Inhibition of cyclooxygenase-1 and 12-lipoxygenase (in vitro). Results: (a) PMA model: pR(-)-[2.2]paracyclophaneacetic acid (10-6 mmol/ear): 58% inhibition after 24 h ( $p < 0.05$ ). (b) Oxazolone model: pR(-)-[2.2]paracyclophaneacetic acid (10-6 mmol/ear): 42% inhibition after 24 h ( $p < 0.05$ ). (c) Carrageenan model: pR(-)-[2.2]paracyclophaneacetic acid (10 mg/kg): 31.4% inhibition (paw volume  $0.48 \pm 0.13$  mL). (d) Cyclooxygenase-1 and 12-lipoxygenase: no inhibition at concns. up to 10  $\mu$ M. Conclusions: The easily accessible [2.2]paracyclophane moiety should find its use in medicinal chemical as it is a pharmacophoric substituent with the interesting feature of planar chirality.

IT 2627-86-3, S-(-)-1-Phenylethylamine 3886-69-9,  
 R-(+)-1-Phenylethylamine  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (antiinflammatory planar chiral [2.2]paracyclophaneacetic acid enantiomers)

RN 2627-86-3 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 1-3 (Pharmacology)  
 IT 2627-86-3, S-(-)-1-Phenylethylamine 3886-69-9,  
 R-(+)-1-Phenylethylamine  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (antiinflammatory planar chiral [2.2]paracyclophaneacetic acid  
 enantiomers)  
 OSC.G 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS RECORD (4  
 CITINGS)  
 RE.CNT 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 11 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN  
 AN 1999:426756 HCAPLUS Full-text  
 DN 131:153202  
 TI Supercritical fluid extraction for selective extraction of  
 enantiomers  
 AU Bauza, Roberto; Rios, Angel; Valcarcel, Miguel  
 CS Analytical Chem. Div., Fac. Sci., Univ. Cordoba, Cordoba, E-14004,  
 Spain  
 SO Analytica Chimica Acta (1999), 391(3), 253-256  
 CODEN: ACACAM; ISSN: 0003-2670  
 PB Elsevier Science B.V.  
 DT Journal  
 LA English  
 AB Selected chiral carboxylic acids (mandelic acid, phenylpropionic acid  
 and phenylbutyric acid) were selectivity extracted from diatomaceous  
 earth with supercrit. CO2 on addition in situ of (R)-(+)- or (S)-(-)-  
 methylbenzylamine as a chiral base. In all cases, a remarkable  
 partial resolution (61-95%) was achieve on an anal. scale. Pressure,  
 temperature and extraction time as well as the mole ratio of base and  
 acid had a marked influence on the quant. extraction of the products.

IT 2627-86-3 3886-69-9  
 RL: ARU (Analytical role, unclassified); MOA (Modifier or  
 additive use); ANST (Analytical study); USES (Uses)  
 (for supercrit. fluid extraction for selective extraction of  
 enantiomeric  
 carboxylic acids)  
 RN 2627-86-3 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS

CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 80-4 (Organic Analytical Chemistry)

Section cross-reference(s): 25

IT 2627-86-3 3886-69-9

RL: ARU (Analytical role, unclassified); MOA (Modifier or additive use); ANST (Analytical study); USES (Uses)

(for supercrit. fluid extraction for selective extraction of enantiomeric

carboxylic acids)

OSC.G 10 THERE ARE 10 CAPLUS RECORDS THAT CITE THIS RECORD (10 CITINGS)

RE.CNT 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 12 OF 12 HCAPLUS COPYRIGHT 2010 ACS on STN

AN 1995:887264 HCAPLUS Full-text

DN 124:41245

OREF 124:7605a,7608a

TI Solute-solvent chiral interactions: non-symmetrical effects of enantiomers and conformers on right- and left-handed cholesterics

AU Yarovoy, Y. K.; Labes, M. M.

CS Dep. Chemistry, Temple University, Philadelphia, PA, 19122, USA  
SO Molecular Crystals and Liquid Crystals Science and Technology,  
Section A: Molecular Crystals and Liquid Crystals (1995), 270,  
101-12

CODEN: MCLCE9; ISSN: 1058-725X

PB Gordon & Breach

DT Journal  
 LA English  
 AB Effective helical twisting powers of a number of enantiomers and achiral conformers were measured in highly twisted cholesteric phases varying in both chemical composition and macroscopic chirality. In certain solute-solvent combinations, and particularly in steroidal solvents, pronounced non-sym. effects of enantiomers and conformers were observed on right- and left-handed cholesterics. Achiral rod-like solutes, which can exist in different conformations, were found to behave as though they have a left-handed helical twisting power in both right- and left-handed short pitch steroidal cholesterics. All effects can be interpreted as being due to specific short range mol.-mol. interactions. No evidence was found that the macroscopic chirality of a cholesteric medium can influence the conformation of an achiral solute.

IT 2627-86-3 3886-69-9  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (solute-solvent chiral interactions and non-sym. effects of enantiomers and conformers on right- and left-handed cholesterics)

RN 2627-86-3 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ S)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (-).



RN 3886-69-9 HCAPLUS  
 CN Benzenemethanamine,  $\alpha$ -methyl-, ( $\alpha$ R)- (CA INDEX NAME)

Absolute stereochemistry. Rotation (+).



CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)  
 Section cross-reference(s): 75

IT 552-79-4 2627-86-3 3886-69-9 5978-70-1,  
 L-2-Octanol 6169-06-8, D-2-Octanol 18434-08-7 18531-94-7,  
 (R)-1,1'-Bi-2-naphthol 18531-99-2, S-1,1'-Bi-2-naphthol  
 42151-56-4 55217-28-2 86503-56-2 87321-20-8 87360-02-9  
 87420-26-6 126659-62-9 133676-09-2 153171-24-5 171624-06-9  
 171624-07-0  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (solute-solvent chiral interactions and non-sym. effects of  
 enantiomers and conformers on right- and left-handed  
 cholesterics)  
 OSC.G 8 THERE ARE 8 CAPLUS RECORDS THAT CITE THIS RECORD (8  
 CITINGS)

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